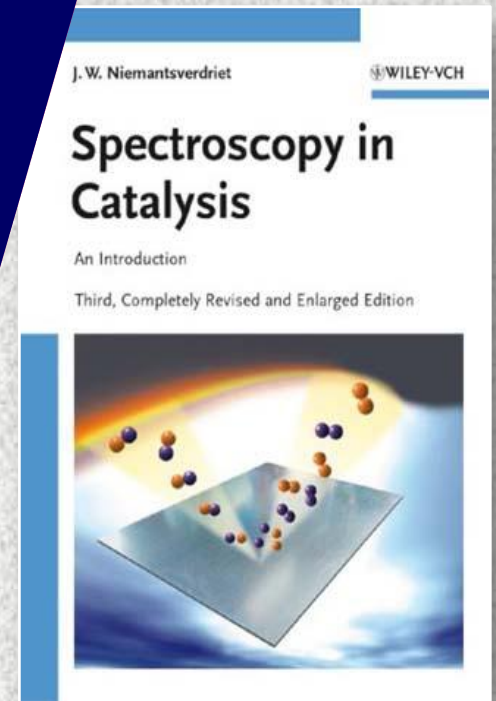


Characterization of solid catalysts

4. Ion Spectroscopy

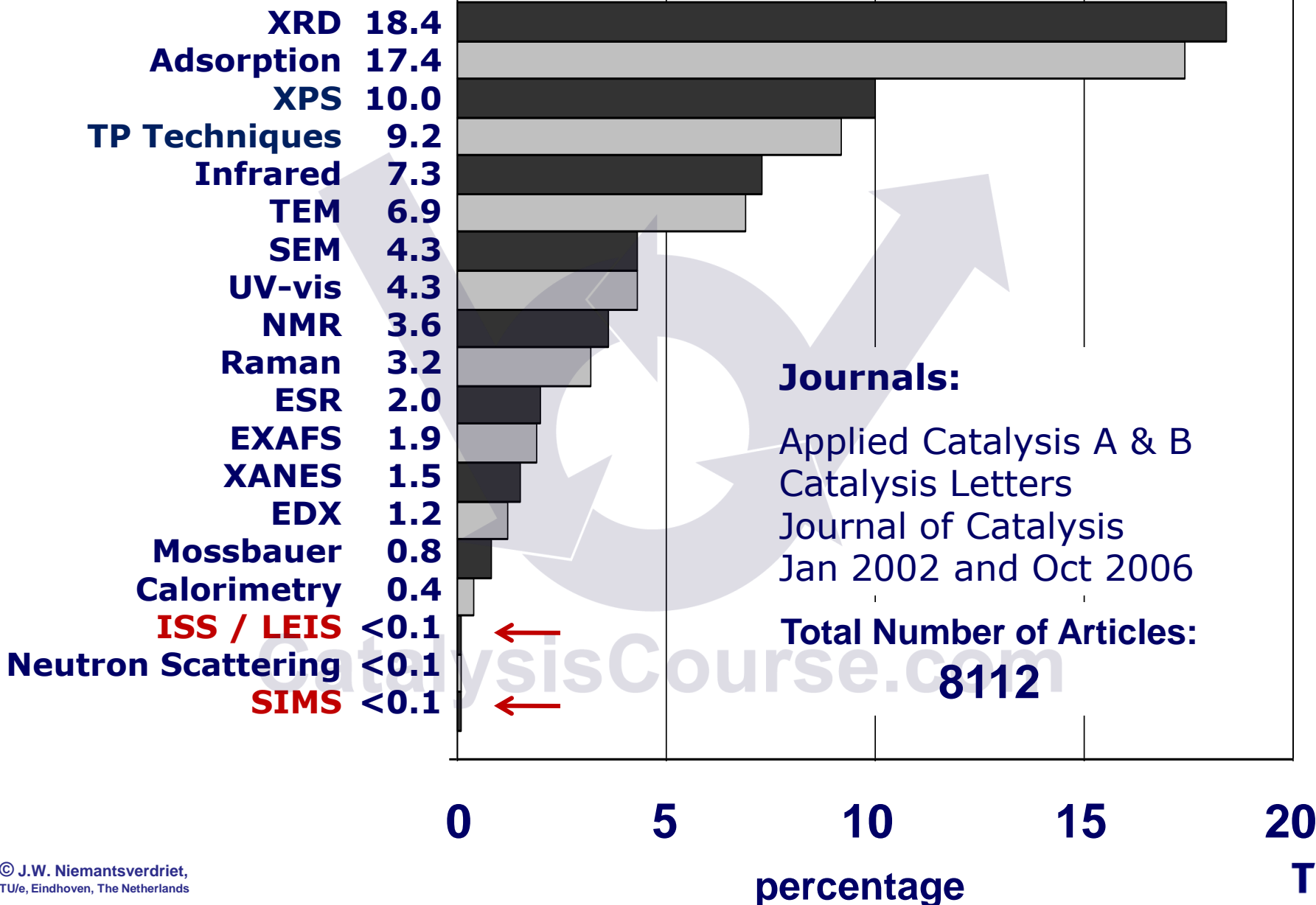
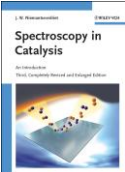
Prof dr J W (Hans) Niemantsverdriet
Schuit Institute of Catalysis

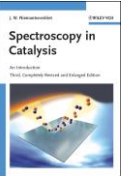


TU / **e**

Technische Universiteit
Eindhoven
University of Technology

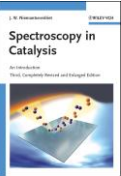
How often are techniques used



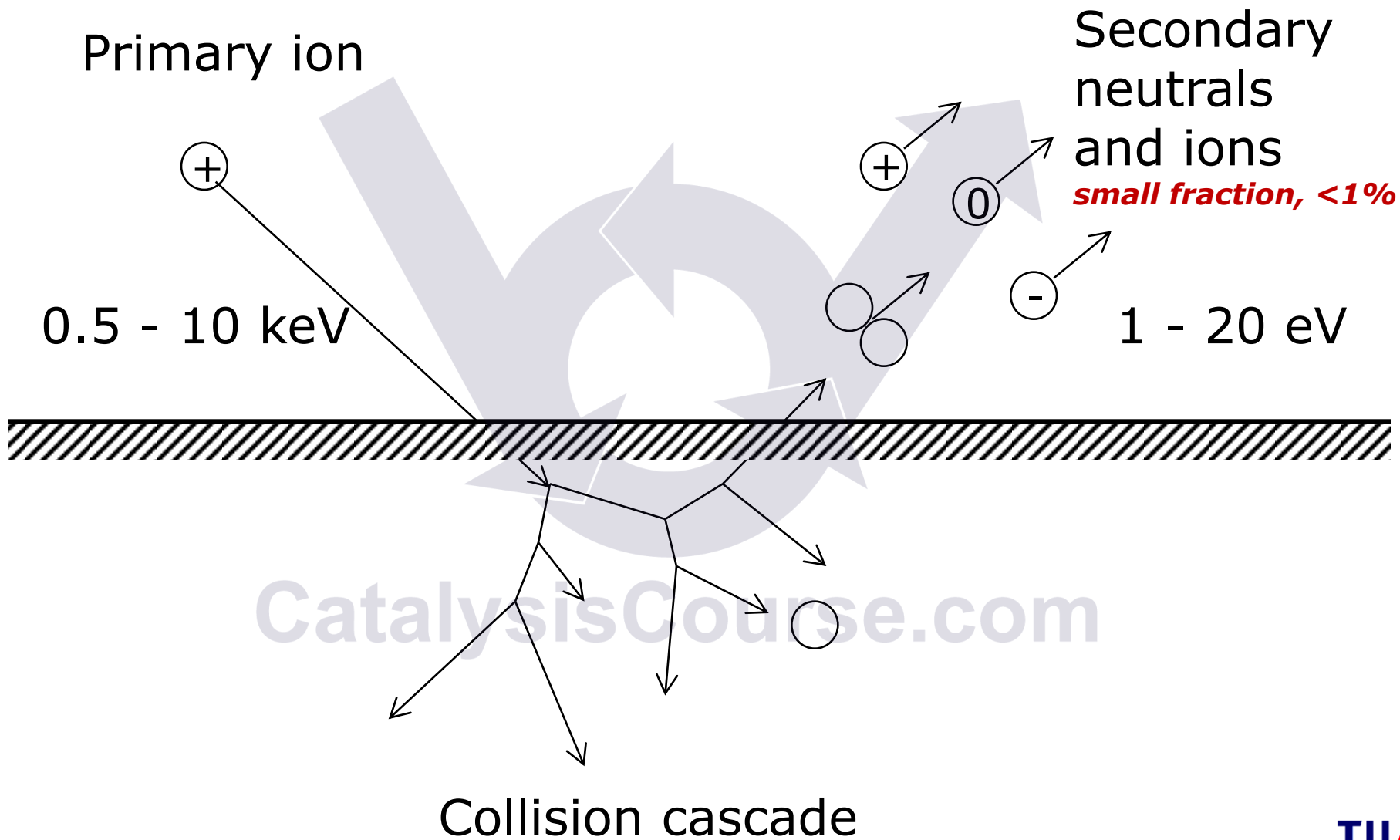


Ion Spectroscopy

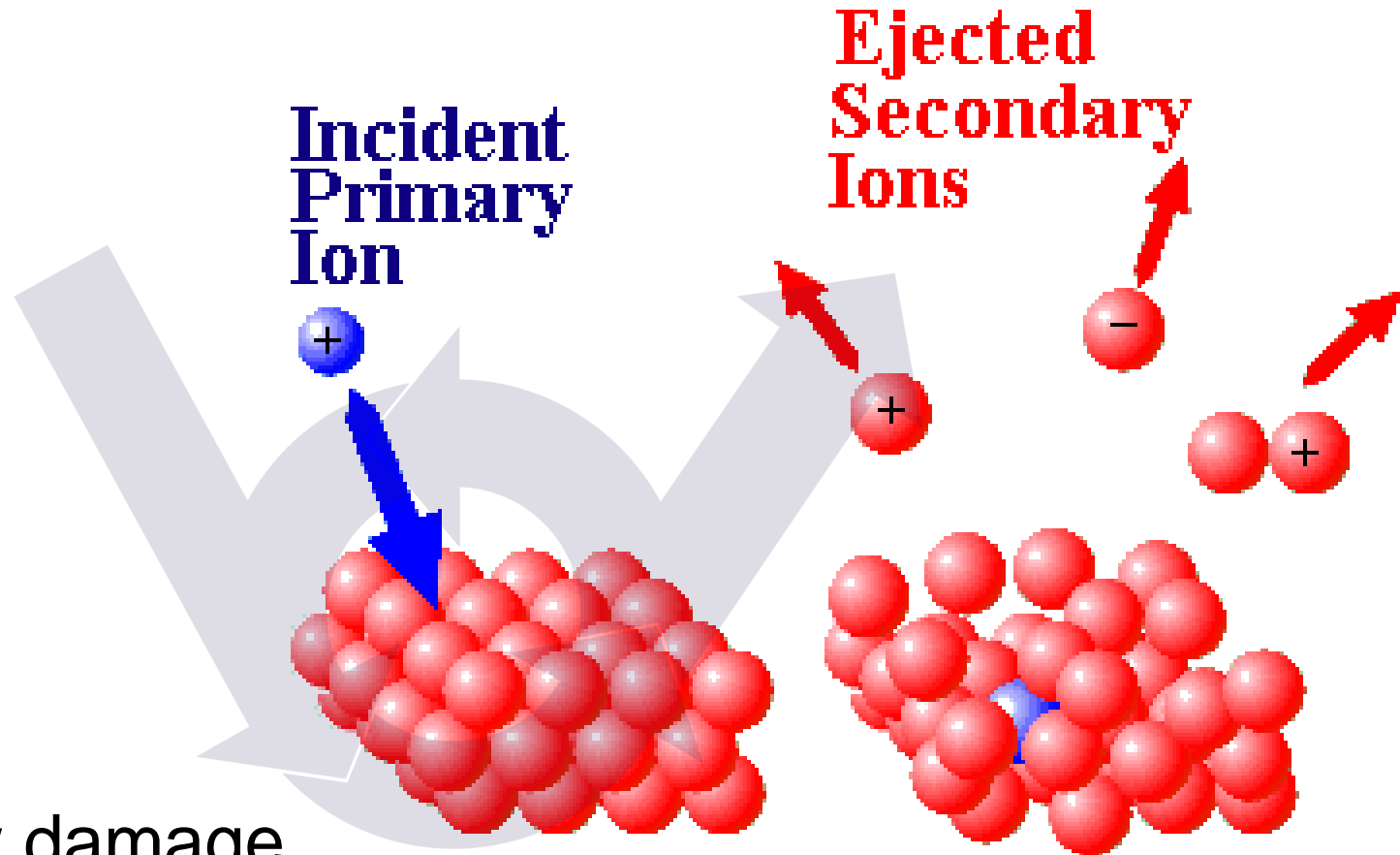
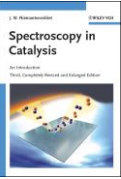
- SIMS secondary ion mass spectrometry
- SNMS secondary neutral mass spectrometry
- ISS ion scattering spectroscopy
- RBS Rutherford backscattering spectroscopy
- LEIS low energy ion spectroscopy



Secondary Ion Mass Spectrometry



Static & Dynamic SIMS



Static = low damage

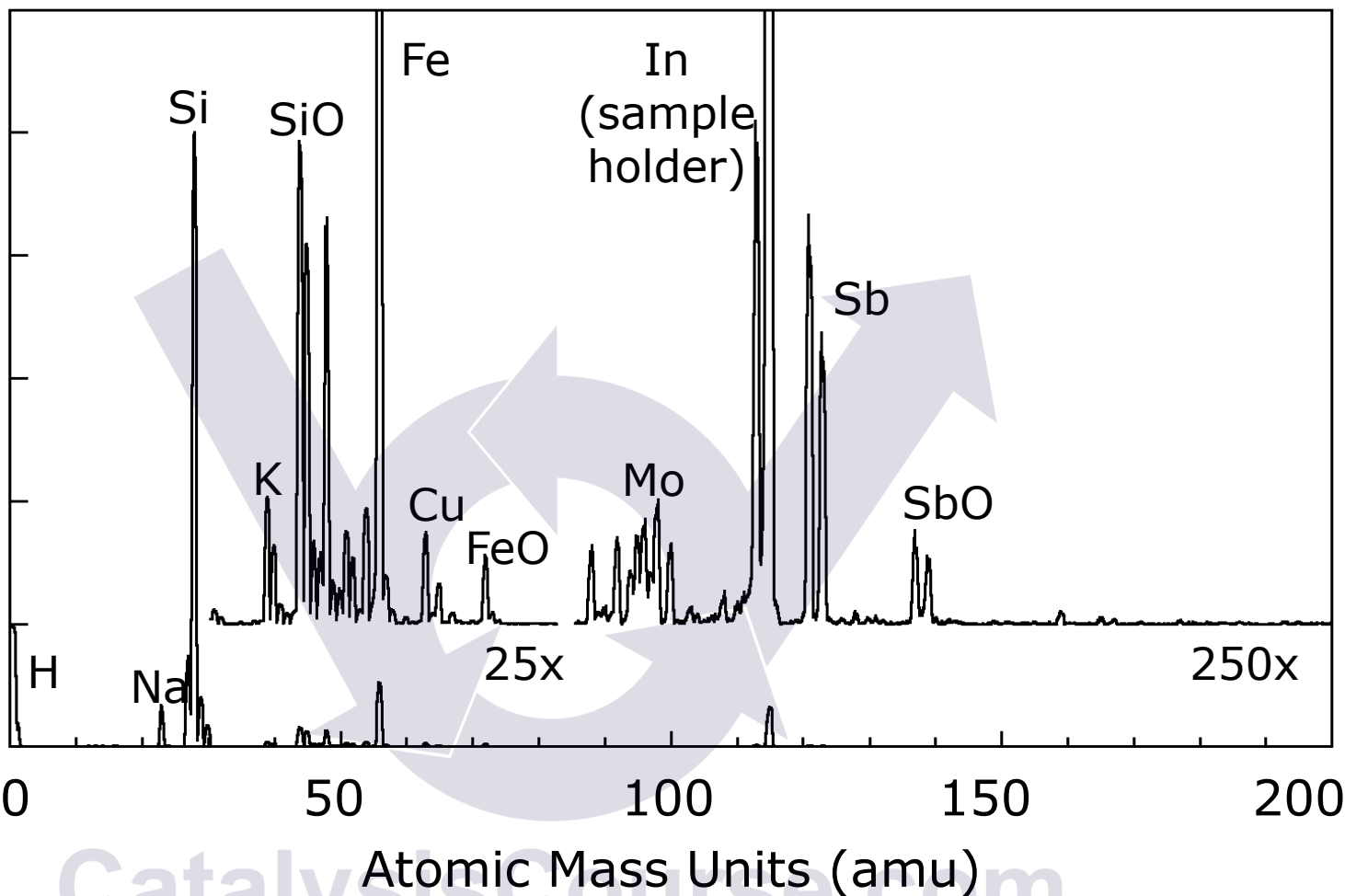
Monolayer removed in minutes

Dynamic = sputtering

Several layers per second removed

Courtesy VG

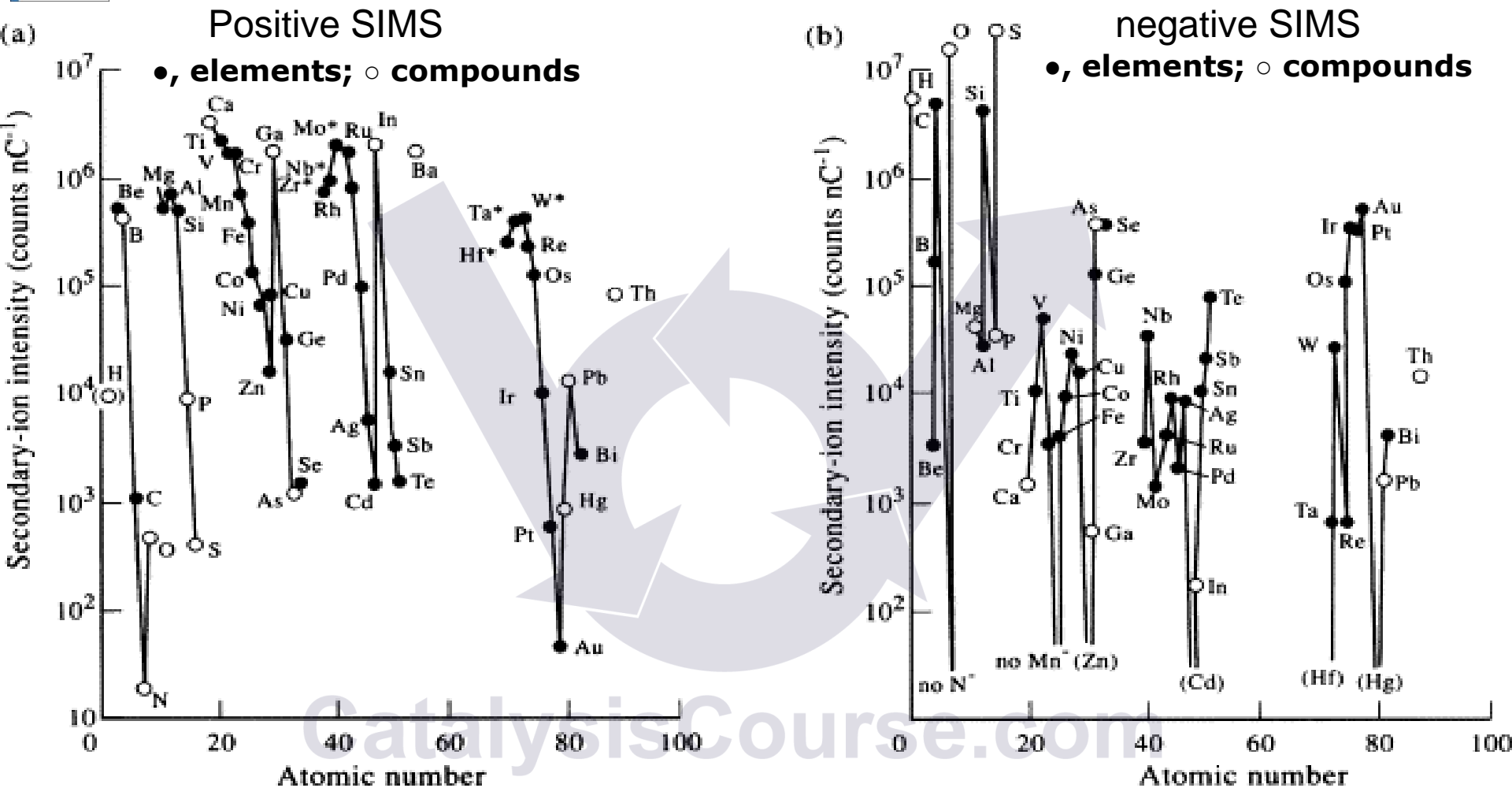
SIMS Fe-Sb-Oxide Ammoxidation Catalyst



Herman Borg &
Pieter Gunter,
TU/e 1990

- 1) SIMS reveals all elements (also trace amounts)**
- 2) Note the important role of isotopes in the identification of species**

SIMS Intensities vary strongly over the periodic table



H.A. Storms, K.F. Brown and J.D. Stein, *Anal. Chem.* **49** (1977) 2023

SIMS intensities also vary strongly with oxidation state and environment; this makes quantification very difficult

SIMS intensity:

$$I_s^\pm = I_p Y R^\pm c^{surf} T$$

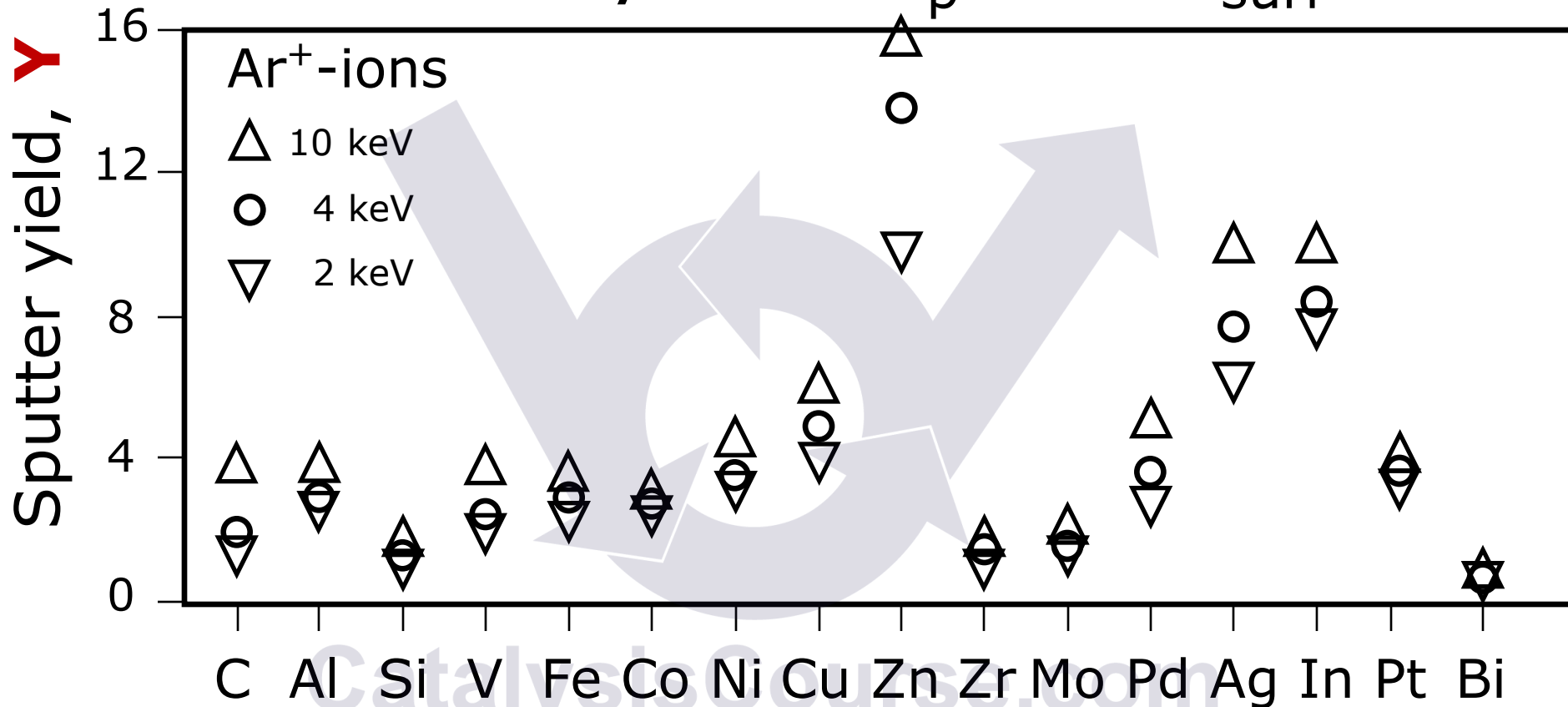
with

- I_s intensity of secondary ions (=rate in counts per sec)
- I_p flux of primary ions
- Y sputter yield (number of atoms ejected per incident ion)
- R probability that particle leaves as positive or negative ion
- c^{surf} fractional concentration of the element in the surface
- T transmission of the mass spectrometer,

(typically 10^{-3} for quadrupole; 10^{-1} for time-of-flight)

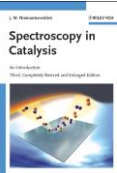
Sputter yields Y are well known

$$\text{Intensity: } I^{\pm} = I_p Y R^{\pm} C_{\text{surf}} T$$



Data from:

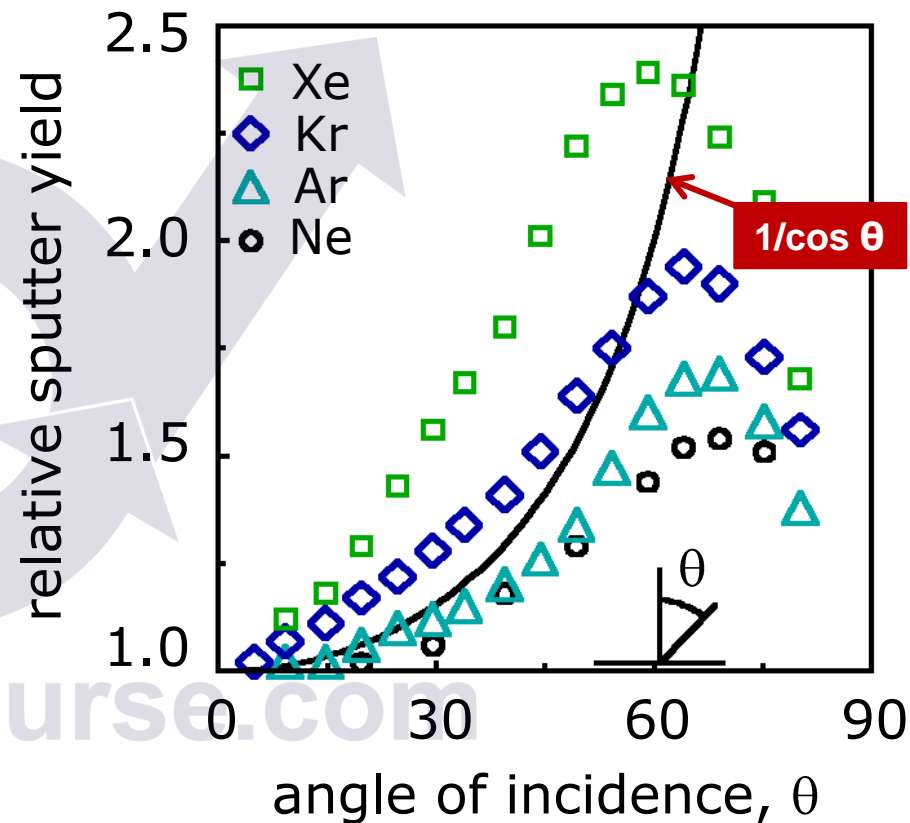
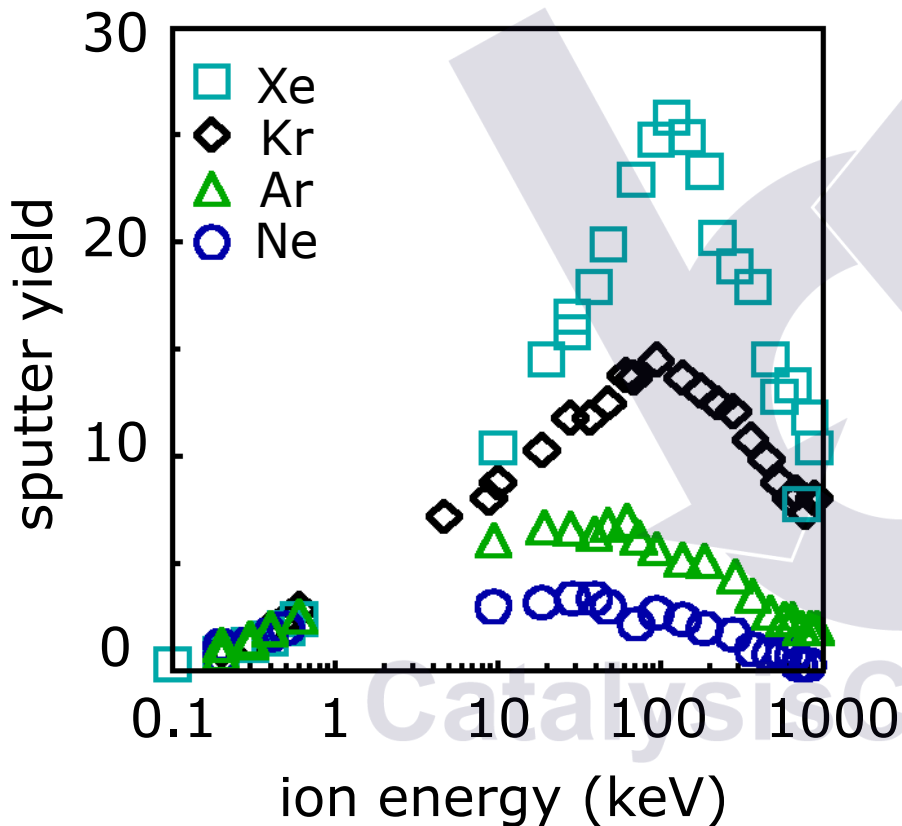
J.C. Vickerman, A. Brown, and N.M. Reed (Eds.),
Secondary Ion Mass Spectrometry, Principles and Applications,
Clarendon, Oxford, 1989.



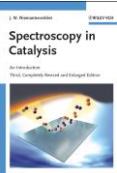
SIMS: Sputter Yields

depend on the primary ion mass and on impact geometry

$$\text{Intensity: } I^{\pm} = I_p \mathbf{Y} R^{\pm} C_{\text{surf}} T$$



Data from: H. Oechsner, Z. Phys. 261 (1973) 37



ionization probability

*in a conceptual, but qualitative model;
depends on probabilities of ionization
and reneutralization upon emission*

$$R^+ \propto e^{const \frac{\phi - I}{v}}$$

R+ ionization probability

φ work function of the sample

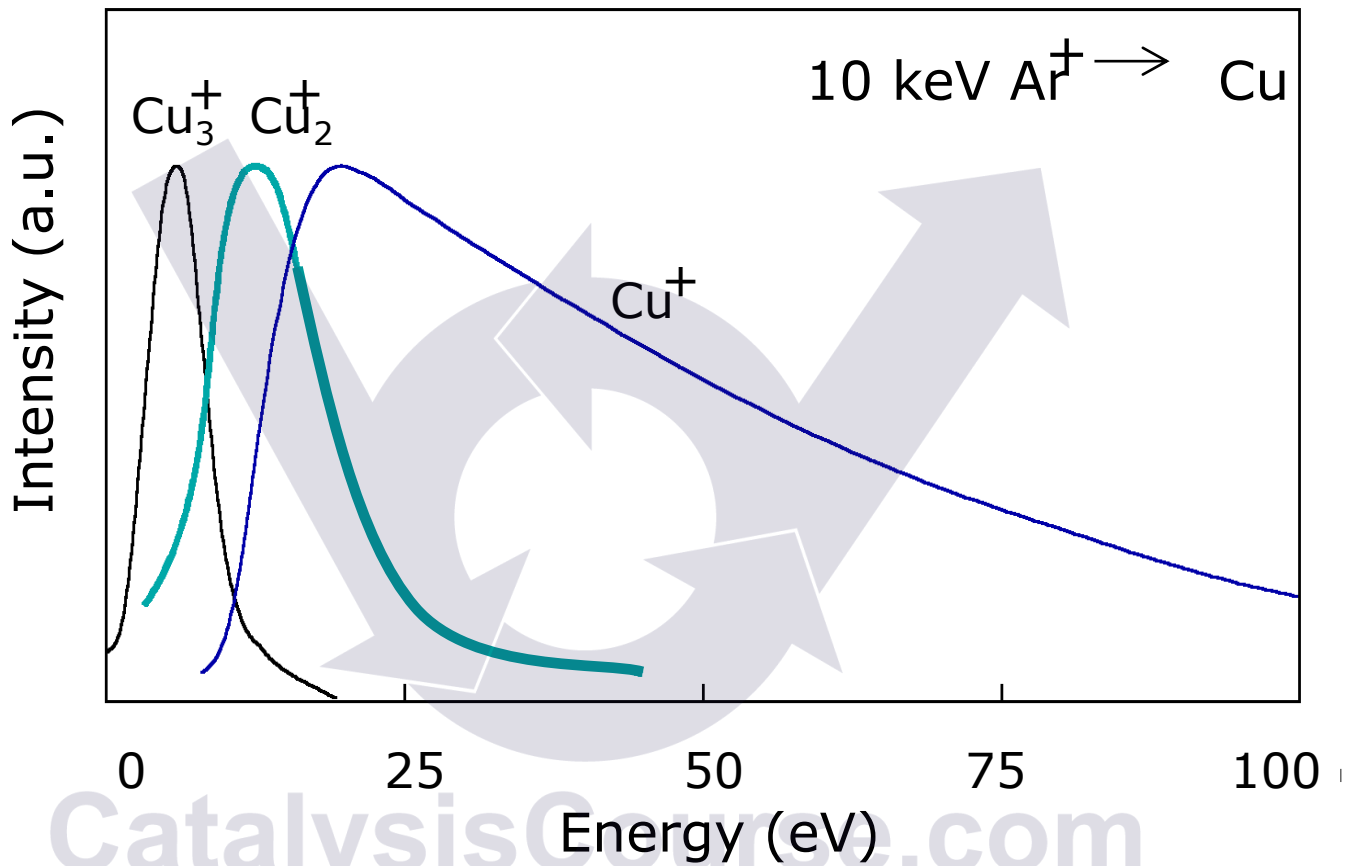
high work function: less chance of neutralization

I ionization potential sputtered particle

low I: higher probability of ionization

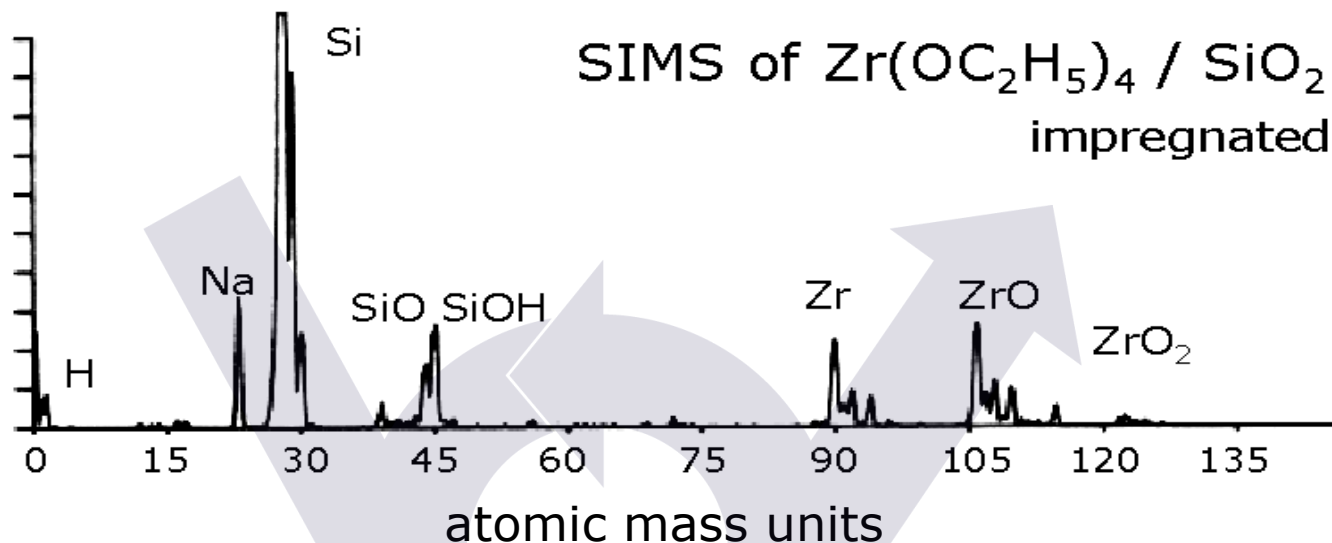
v velocity of the emitted particle

SIMS: Energy Distribution of Secondaries



Implications for charging!

SIMS: Information in Relative Intensities

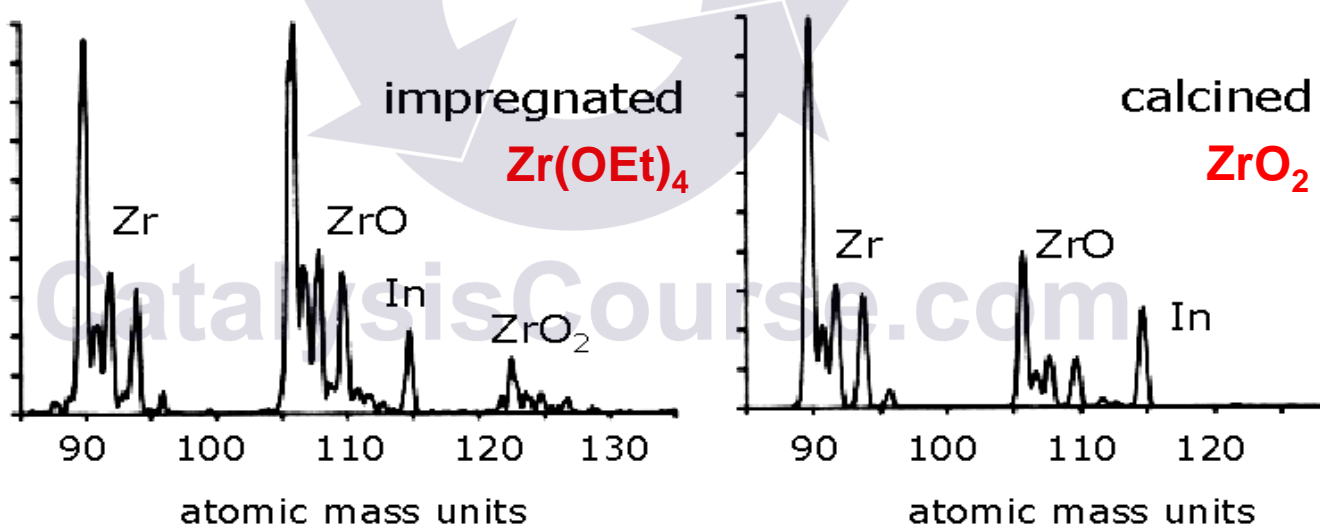
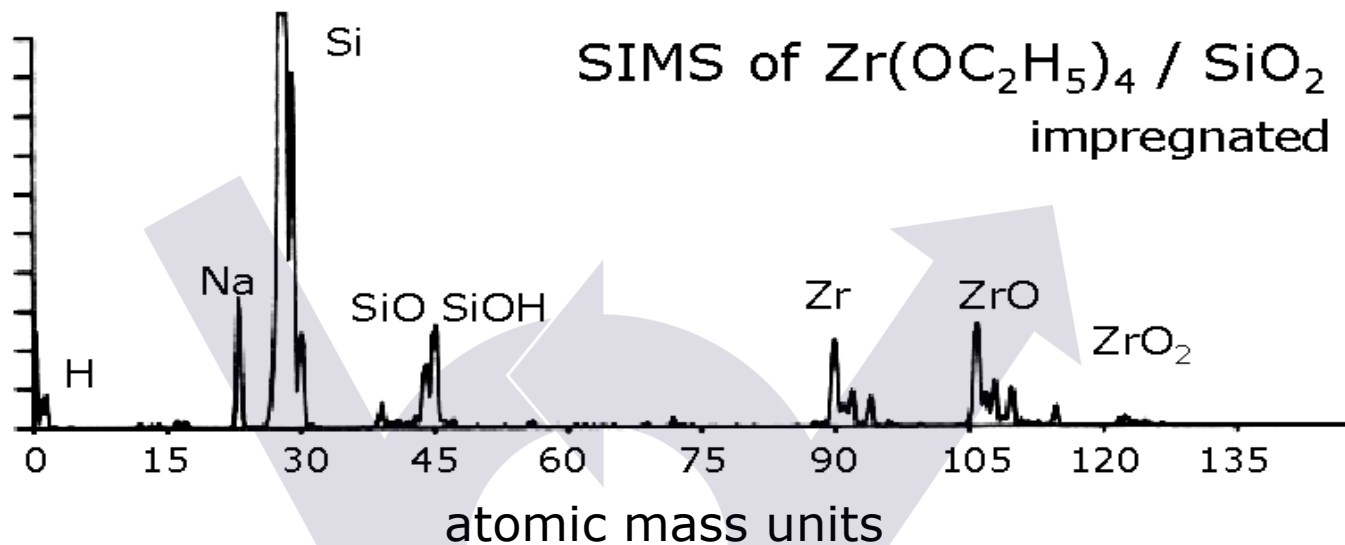


Relative Intensities Zr – ZrO – ZrO₂

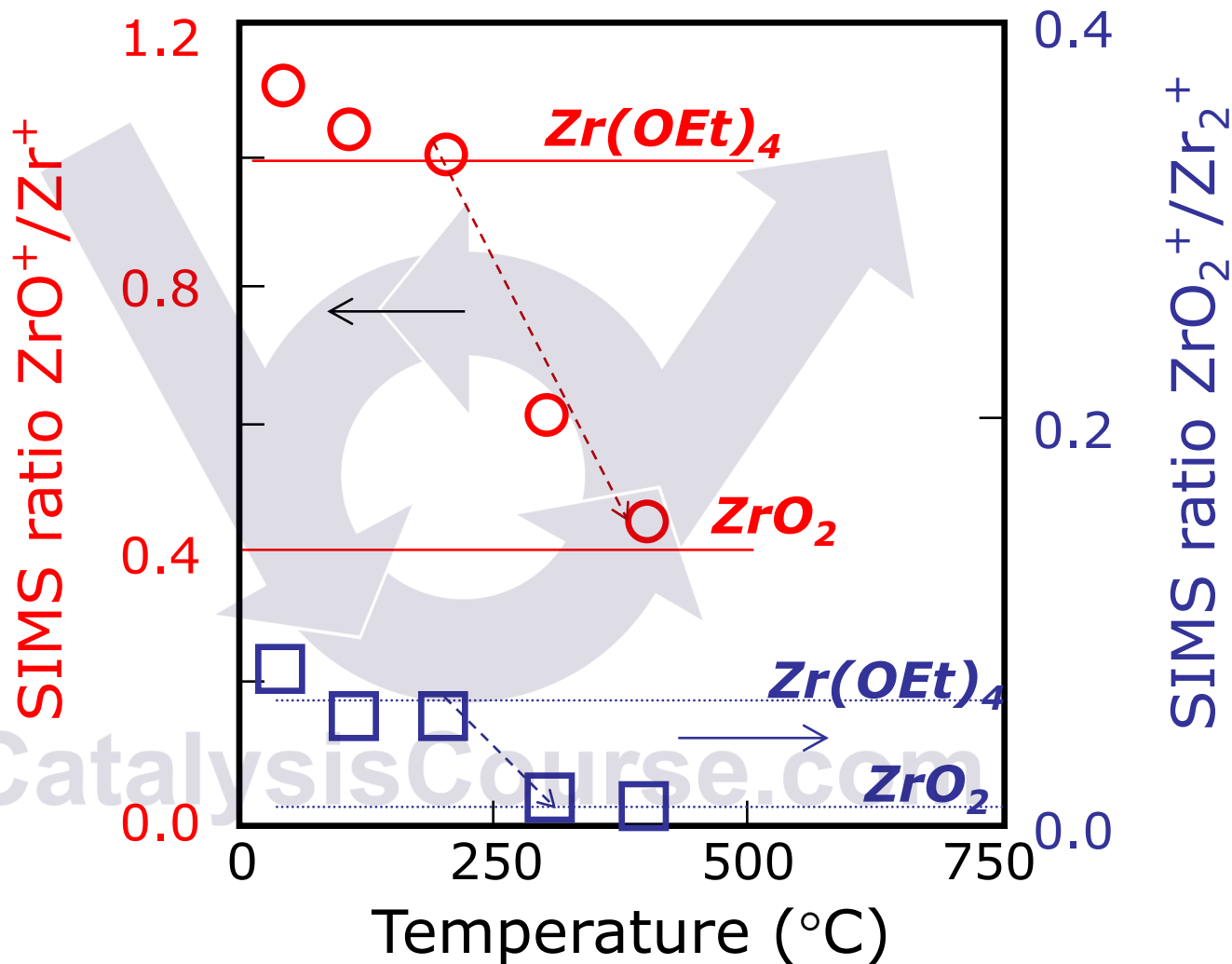
reflect coordination around Zr:

CatalysisCourse.com

SIMS: Information in Relative Intensities



SIMS Reveals Conversion $\text{Zr}(\text{OEt})_4$ to ZrO_2 versus calcination temperature

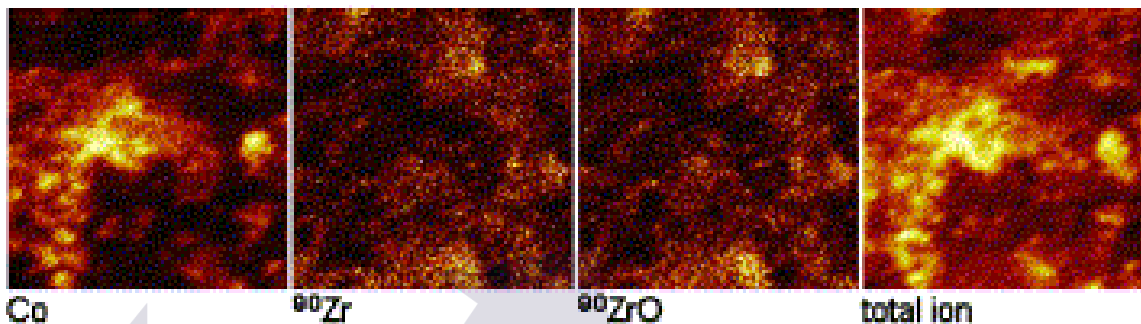


Imaging by SIMS: Chemical Maps

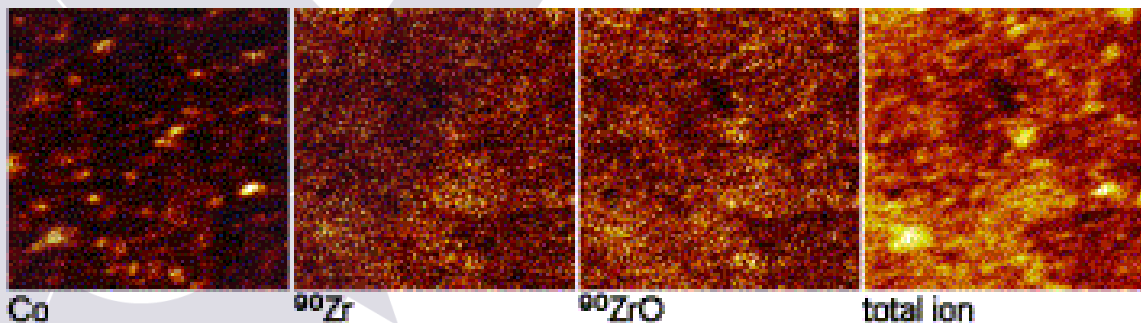
TOF-SIMS Co/ZrO₂ Catalysts

TOF-SIMS high resolution images of 9.9%Co/ZrO₂ catalyst calcined at 350 °C for 3 h. Bright colour indicates investigated ions. Images collected from different surface areas.

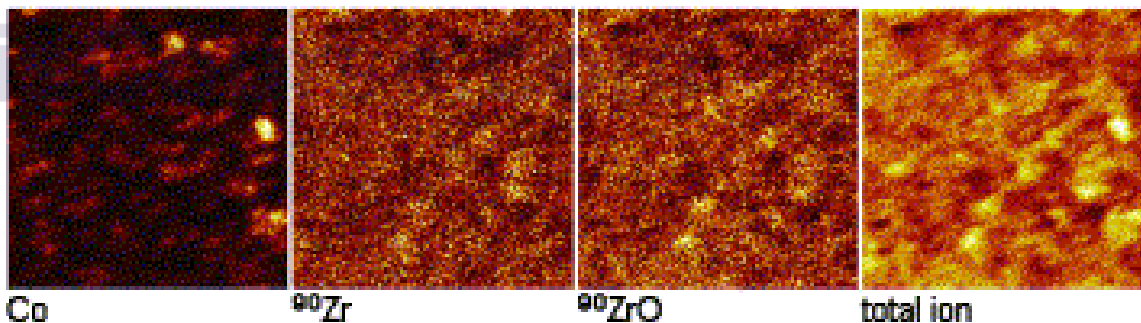
Field of view: 20.5 x 20.5 μm²

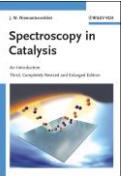


Field of view: 20.5 x 20.5 μm²



Field of view: 21.5 x 21.5 μm²





Ion Spectroscopy

SIMS secondary ion mass spectrometry

SNMS secondary neutral mass spectrometry

ISS ion scattering spectroscopy

RBS Rutherford backscattering spectroscopy

LEIS low energy ion spectroscopy

CatalysisCourse.com

SNMS intensity:

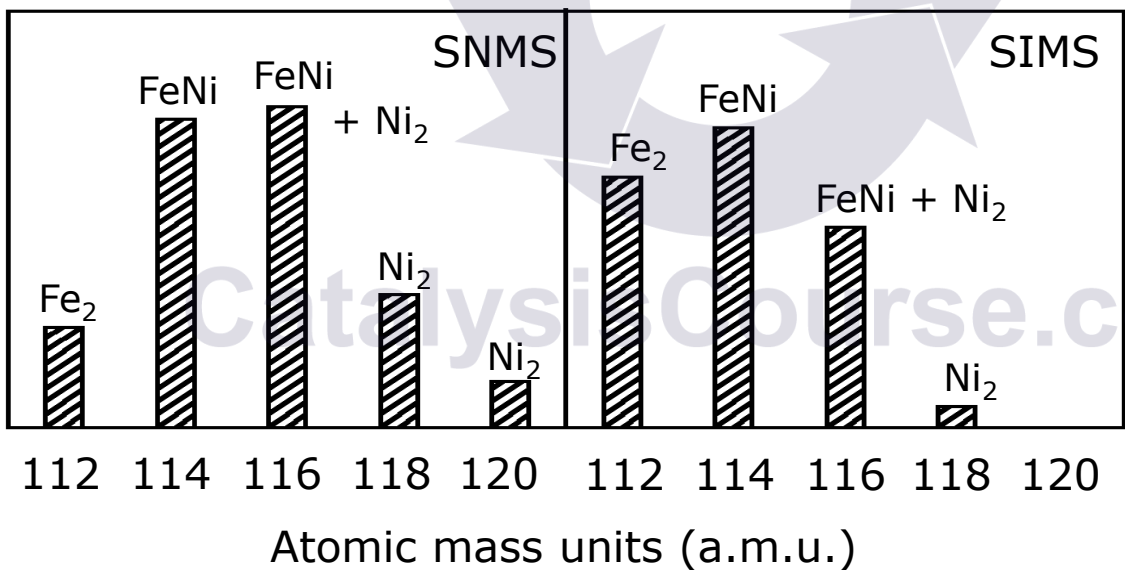
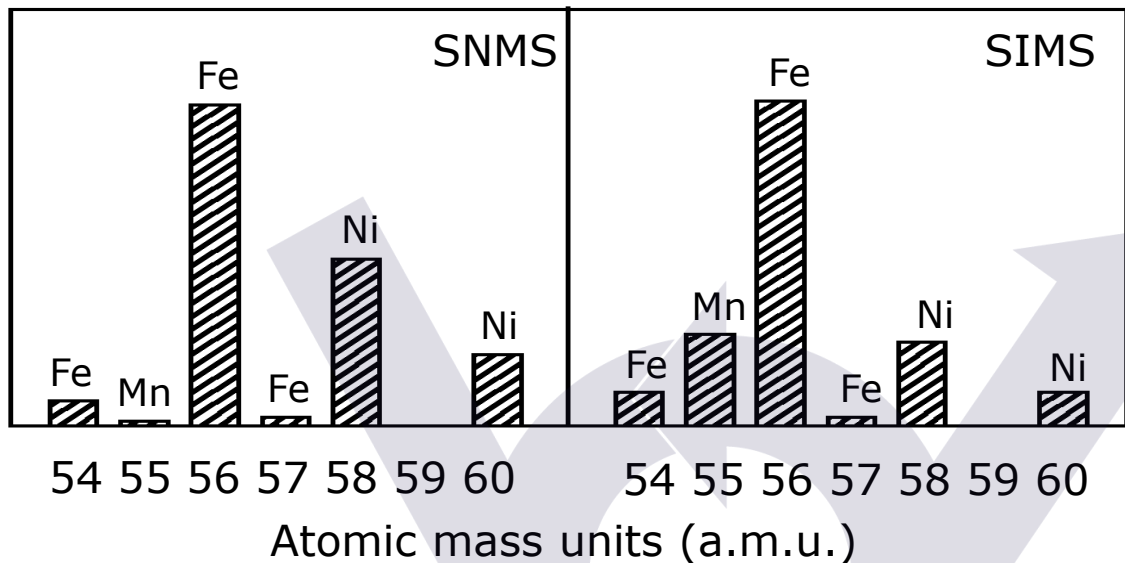
$$I_s^\pm = I_p Y R^\pm c^{surf} T$$

~~$R^\pm \approx 1$~~

with

- I_s intensity of secondary ions (=rate in counts per sec)
- I_p flux of primary ions
- Y sputter yield (number of atoms ejected per incident ion)
- c^{surf} fractional concentration of the element in the surface
- T ionization efficiency of the mass spectrometer (**low!!**)

SIMS & SNMS 1:1 Fe-Ni Alloy



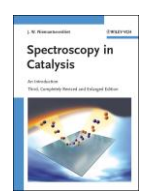
SNMS better quantifiable, but relatively insensitive

Ion Spectroscopy: SIMS

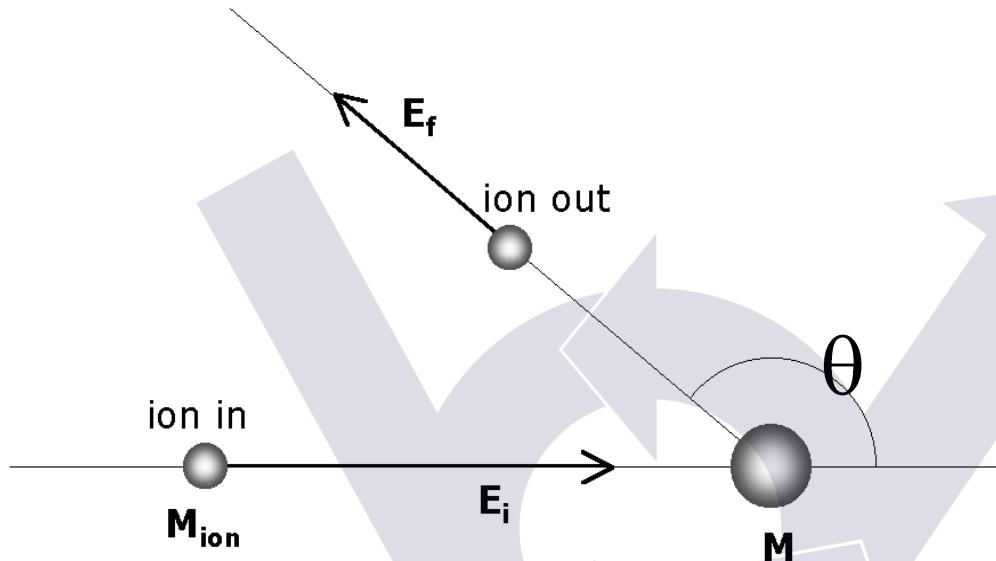
SIMS:

- detection of trace amounts
- molecular information via clusters
- quantification difficult

CatalysisCourse.com



Ion Scattering Spectroscopy

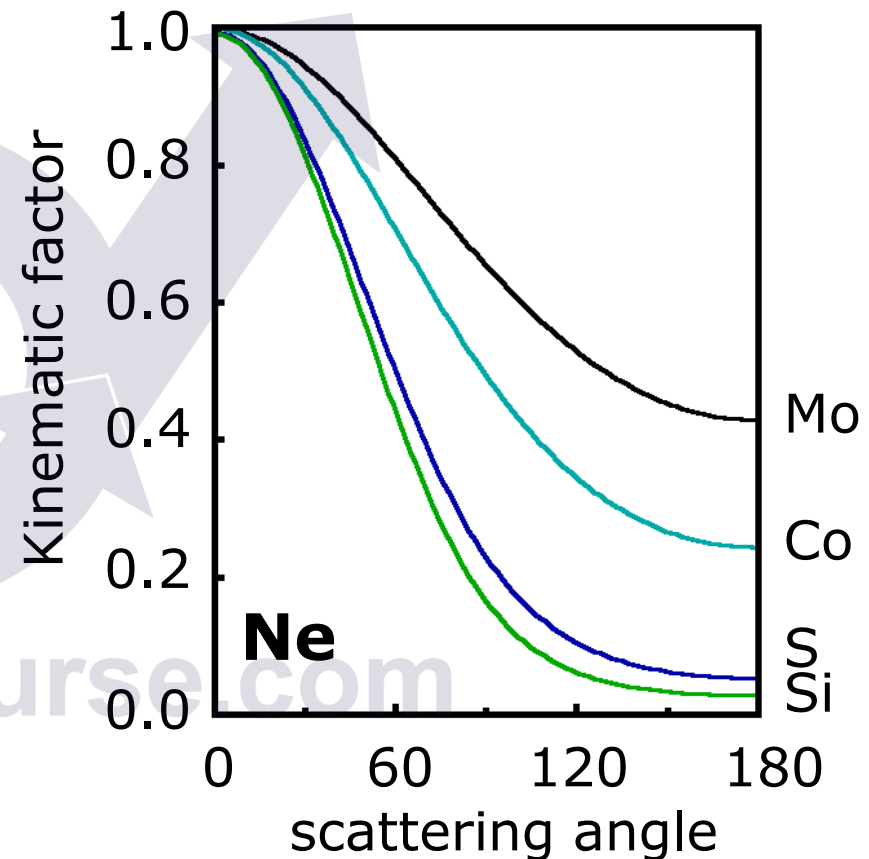
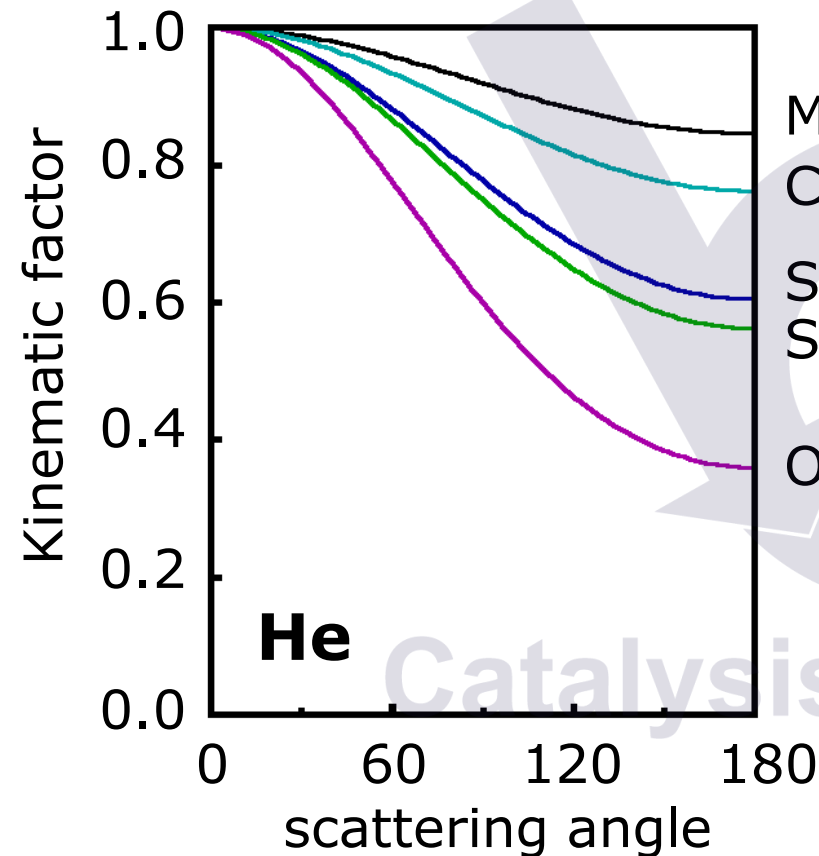


$$K_M = \frac{E_f}{E_i} = \left(\frac{(M^2 - M_{ion}^2 \sin^2 \theta)^{1/2} + M_{ion} \cos \theta}{M + M_{ion}} \right)^2$$

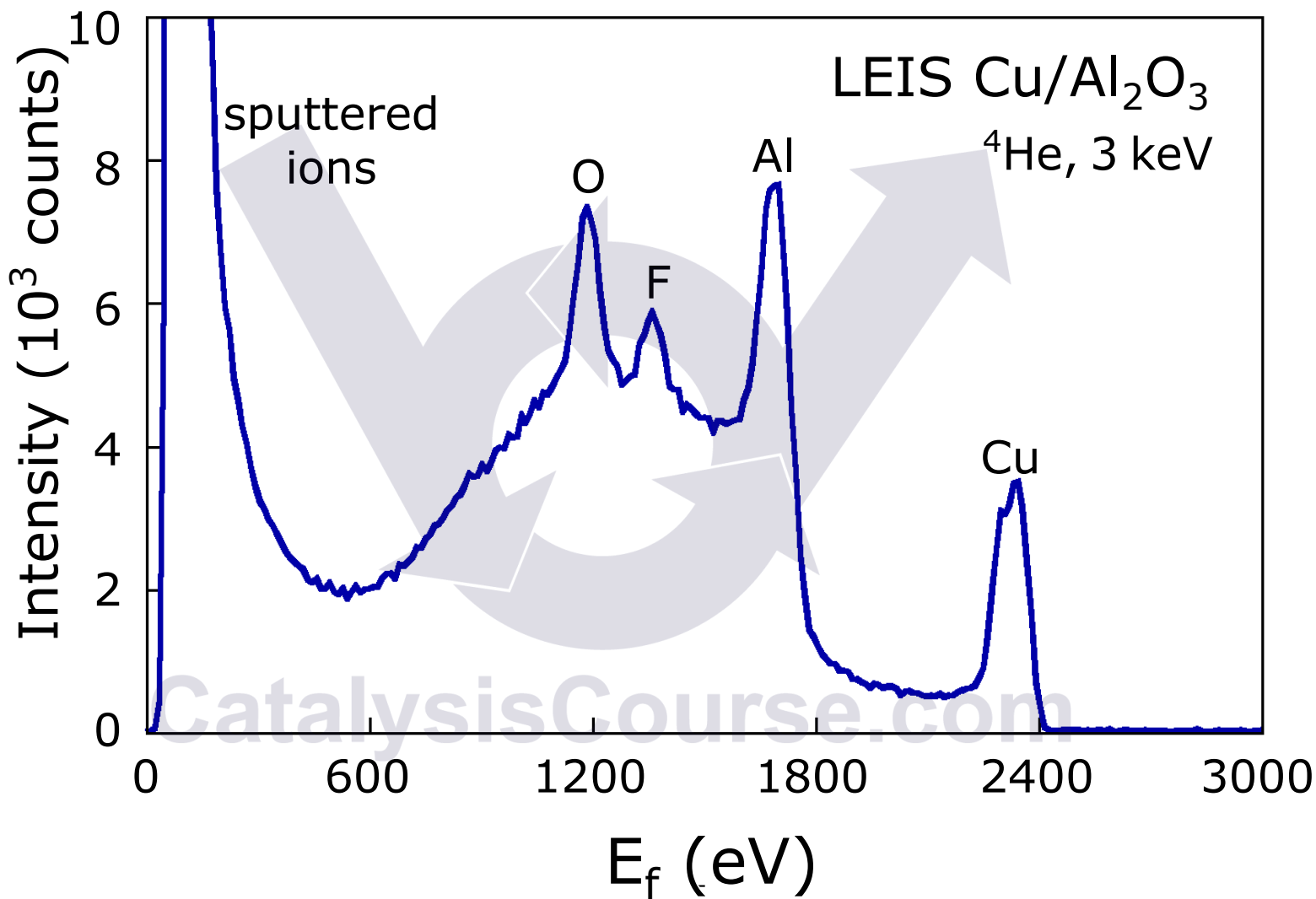
$E_f = \text{Mass Spectrum !!}$

Ion Scattering & Energy Resolution

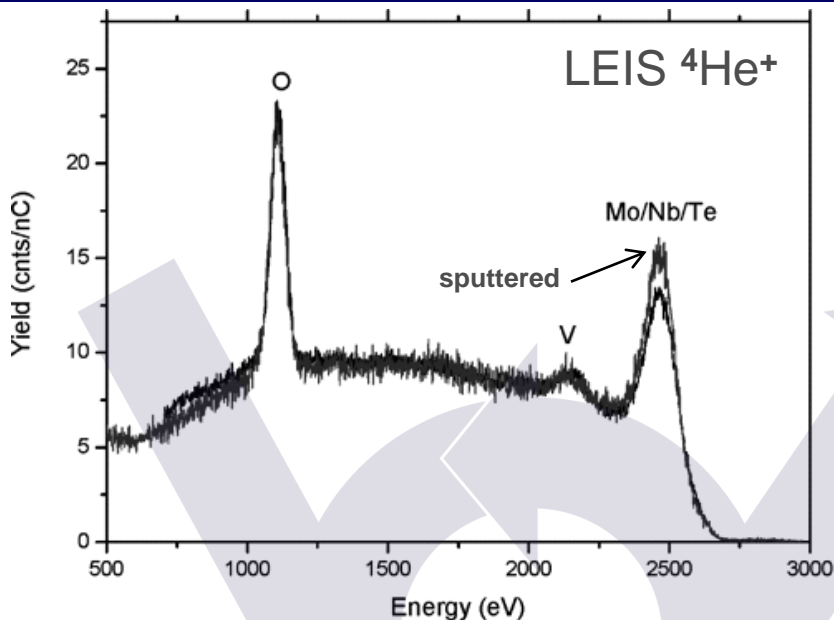
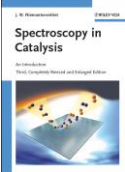
$$K_M = \frac{E_f}{E_i} = \left(\frac{(M^2 - M_{ion}^2 \sin^2 \theta)^{1/2} + M_{ion} \cos \theta}{M + M_{ion}} \right)^2$$



Low Energy Ion Scattering

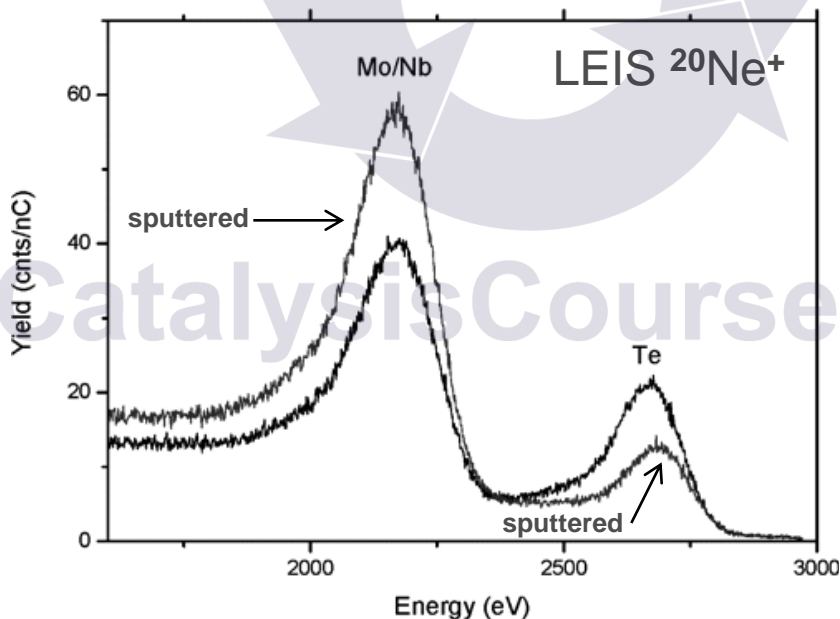


LEIS of V-Mo-Nb-Te mixed metal oxide catalyst



Z		M
8	O	16.00
23	V	50.94
41	Nb	92.90
42	Mo	95.94
52	Te	127.60

Ne scattering:
Better resolution
for higher mass



Light sputtering preferentially
removes Tellurium

V.V. Guliants, R. Bhandari, A.R. Hughett, S. Bhatt, B.D. Schuler, H.H. Brongersma, A. Knoester, A.M. Gaffney, and S. Han, J. Phys. Chem. B 110 (2006) 6129

Ion Spectroscopy

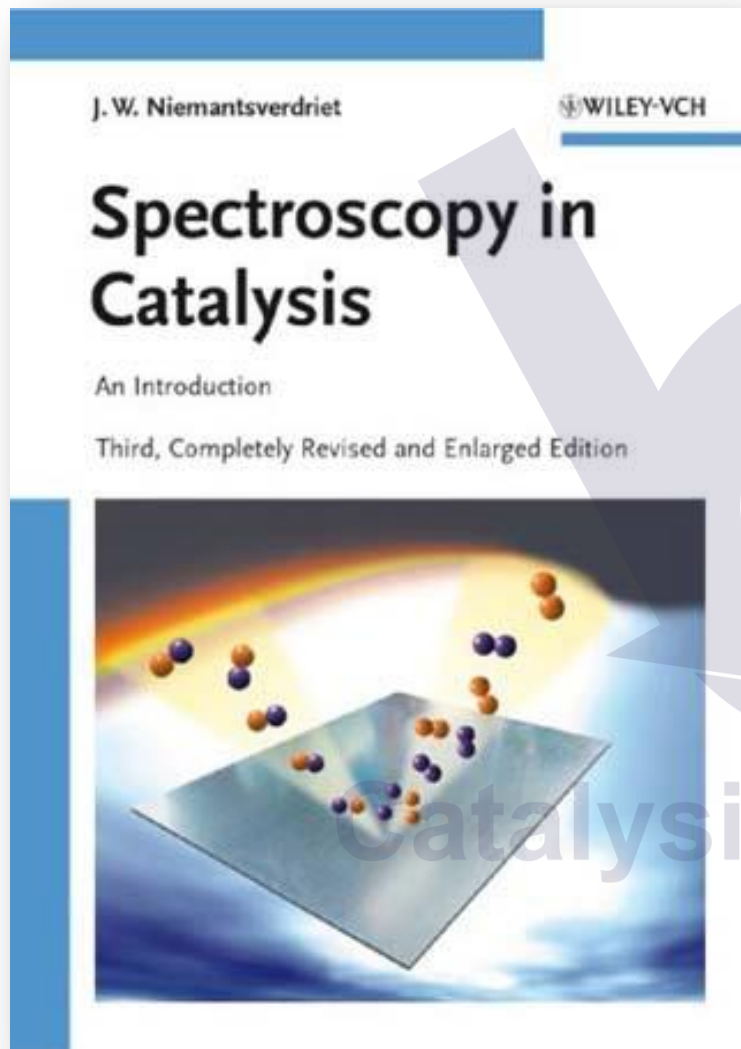
SIMS:

- detection of trace amounts
- molecular information via clusters
- quantitation difficult

LEIS

- extreme surface sensitivity
- reasonably quantitative

Download the handout for this lecture from
www.catalysiscourse.com



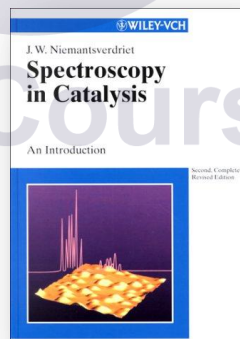
Read more about the ion spectroscopies

in Chapter 4 of

Spectroscopy in Catalysis: An Introduction, Third Edition

J. W. Niemantsverdriet

**Copyright 2007 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim
ISBN: 978-3-527-31651-9**



Version 2000

gives many examples and references to the literature

TU/e Technische Universiteit
Eindhoven
University of Technology