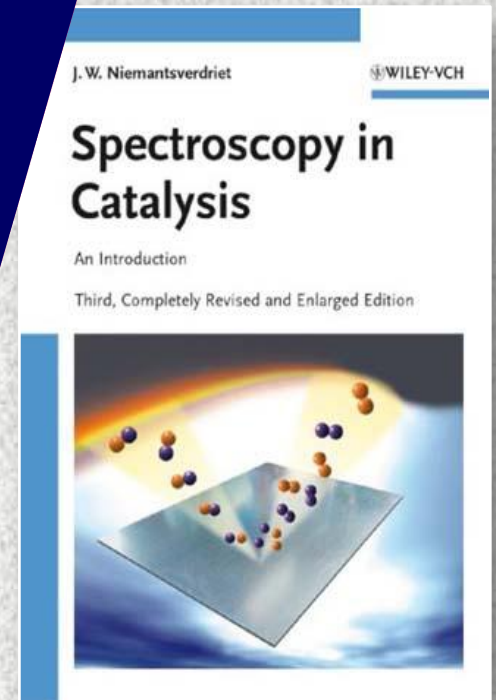


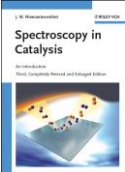
Characterization of solid catalysts

6. X-ray Diffraction

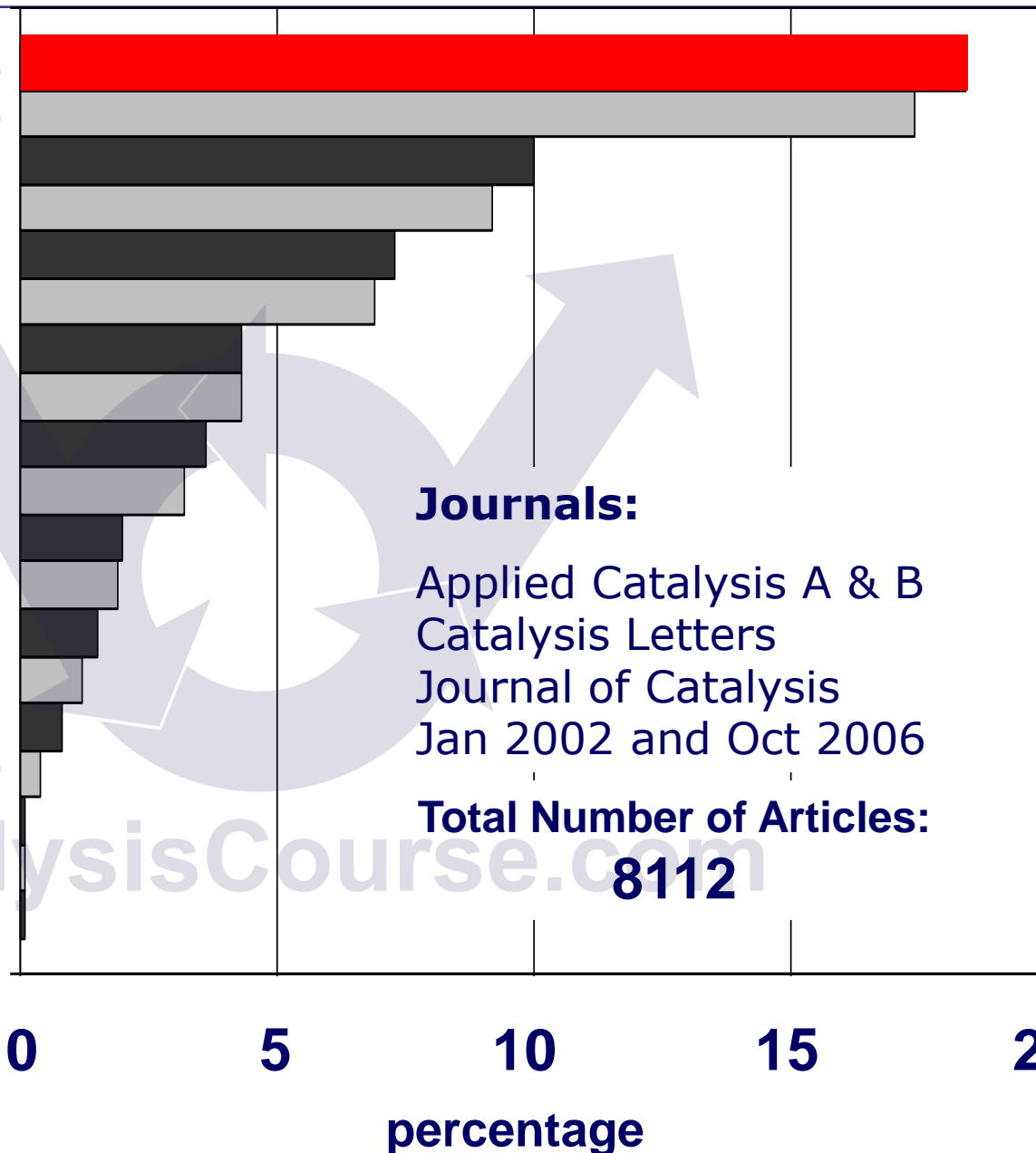
Prof dr J W (Hans) Niemantsverdriet
Schuit Institute of Catalysis



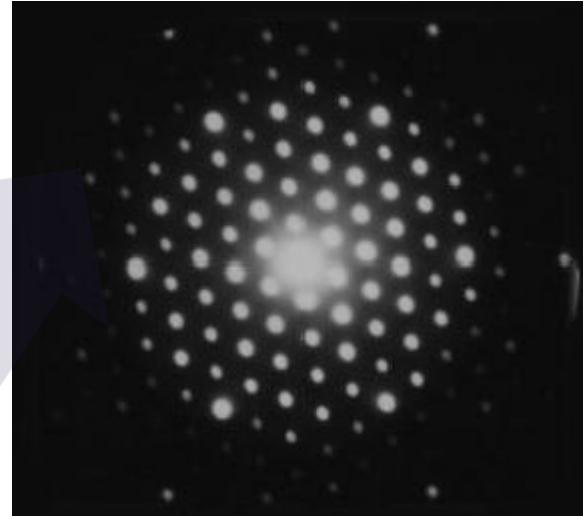
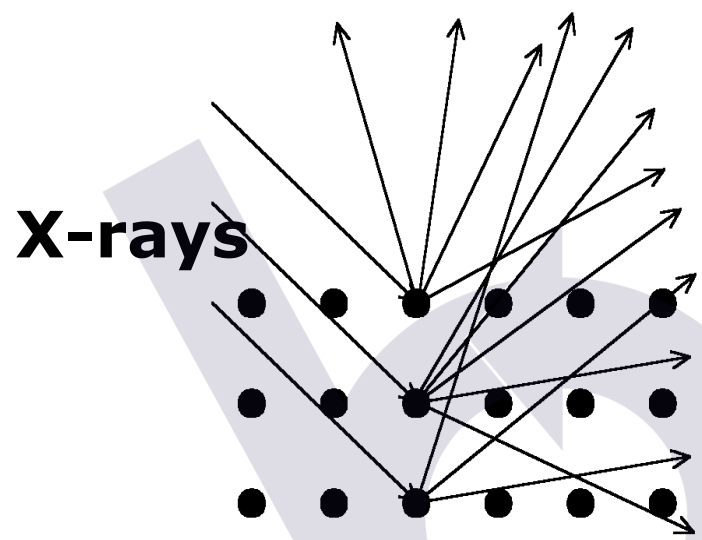
How often are techniques used



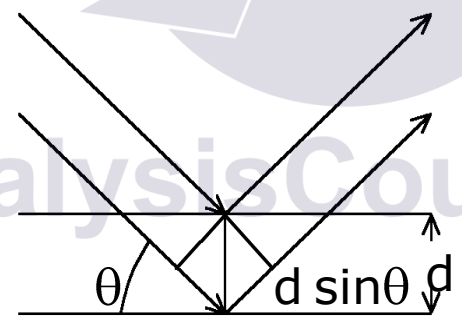
XRD	18.4
Adsorption	17.4
XPS	10.0
TP Techniques	9.2
Infrared	7.3
TEM	6.9
SEM	4.3
UV-vis	4.3
NMR	3.6
Raman	3.2
ESR	2.0
EXAFS	1.9
XANES	1.5
EDX	1.2
Mossbauer	0.8
Calorimetry	0.4
ISS / LEIS	<0.1
Neutron Scattering	<0.1
SIMS	<0.1



XRD: X-ray Diffraction

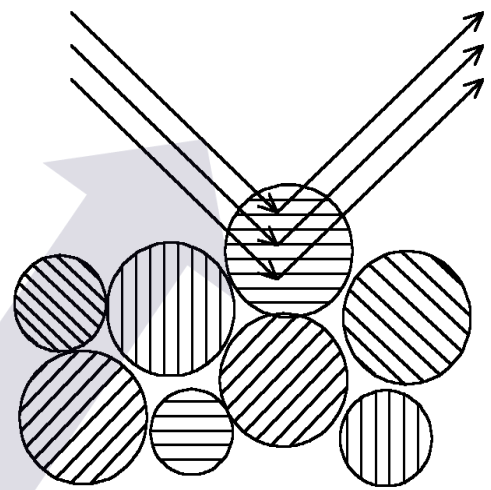
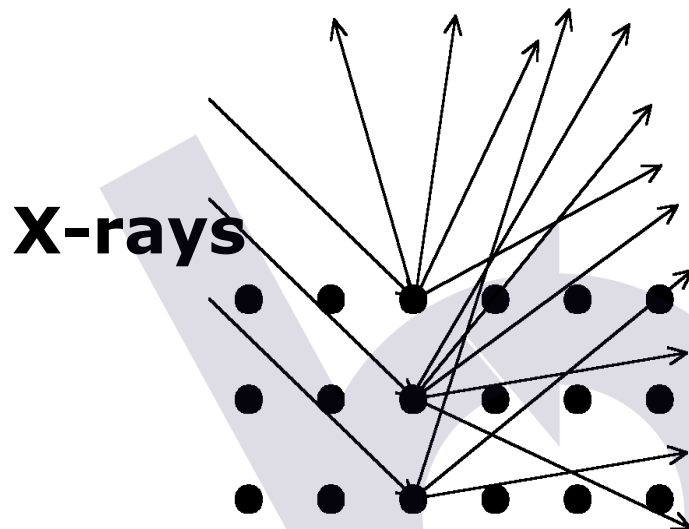


Bragg's Law



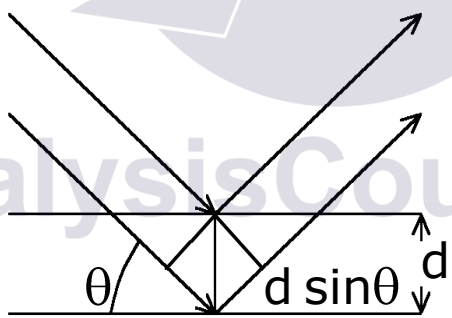
$$n\lambda = 2 d \sin\theta$$

XRD: X-ray Diffraction

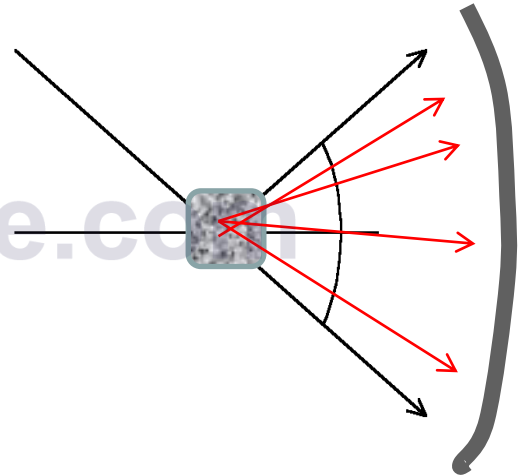


randomly oriented particles

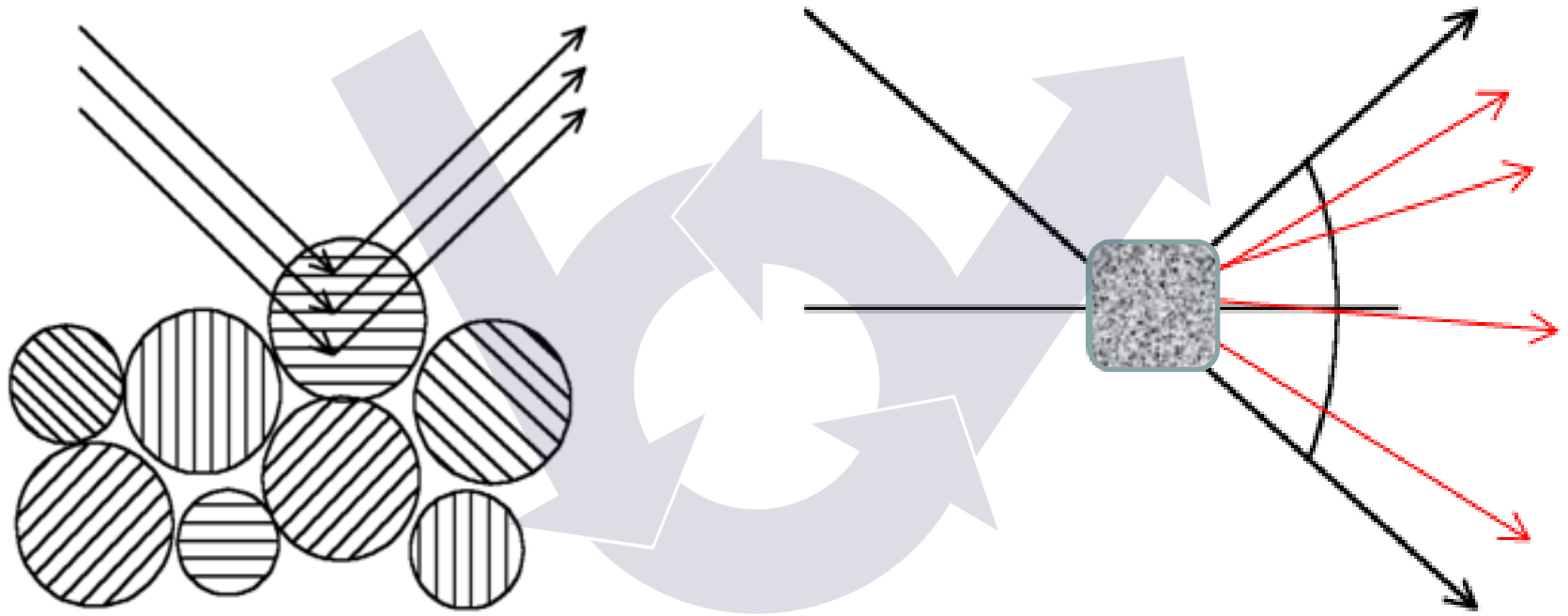
Bragg's Law



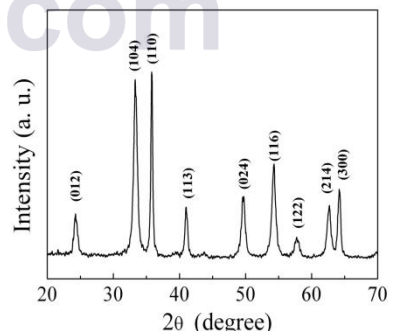
$$n\lambda = 2 d \sin\theta$$



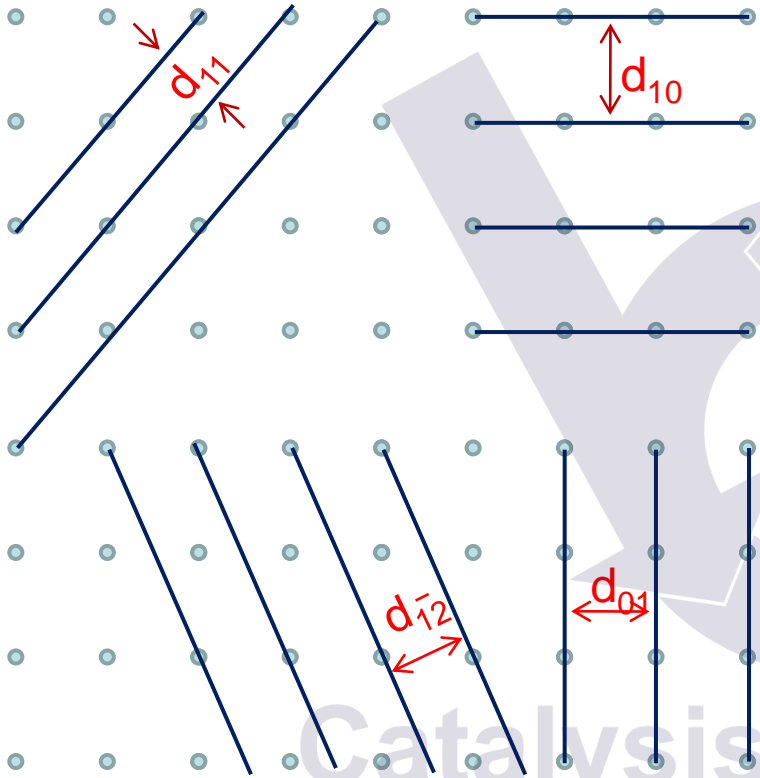
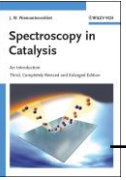
XRD of polycrystalline particles



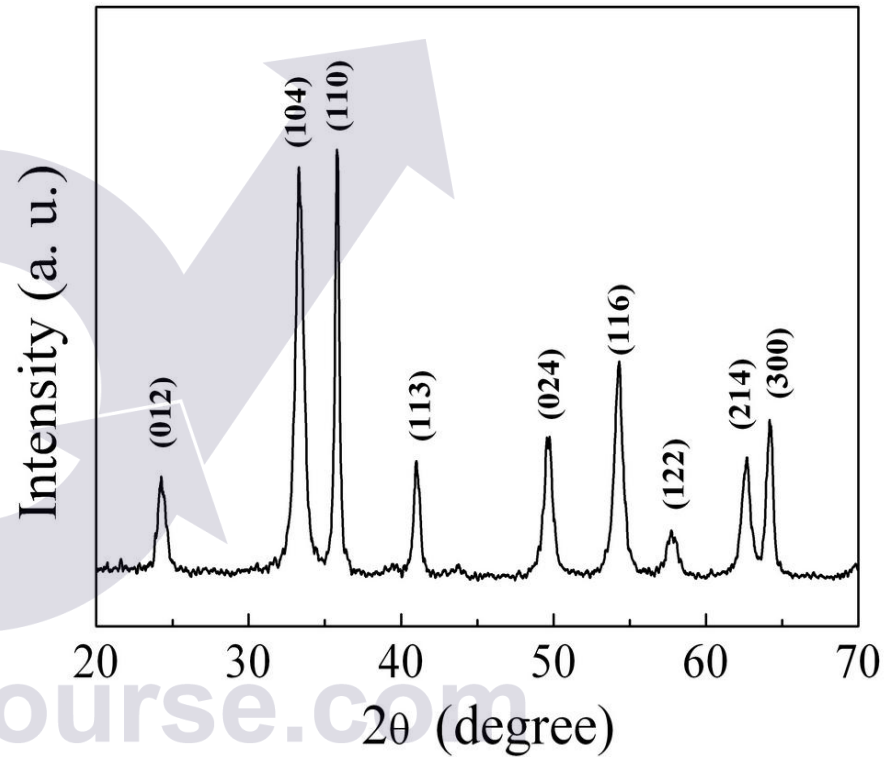
randomly oriented particles



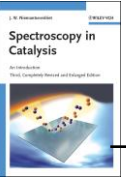
One phase, many lattice spacings



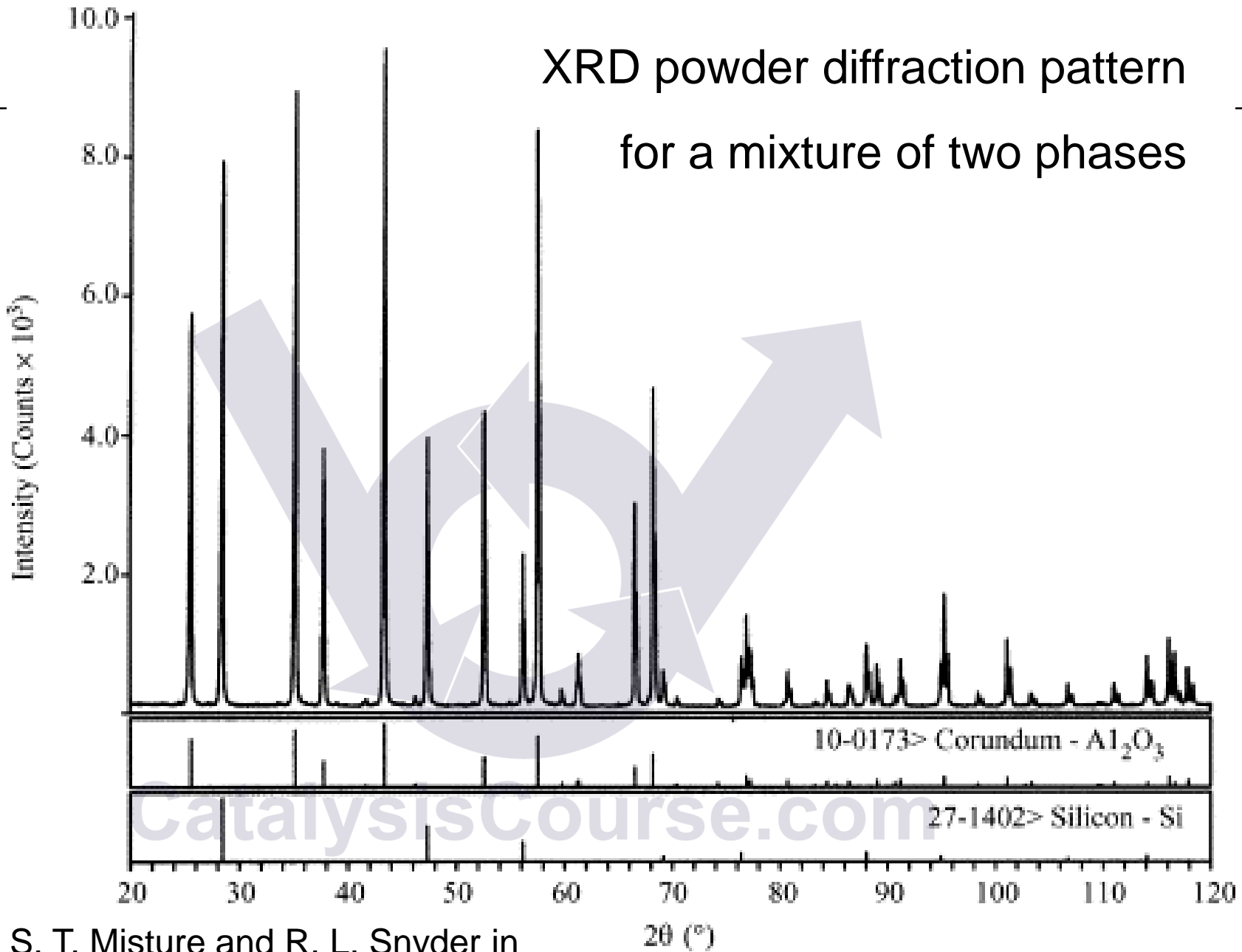
XRD pattern of Fe_2O_3



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XRD powder diffraction pattern for a mixture of two phases



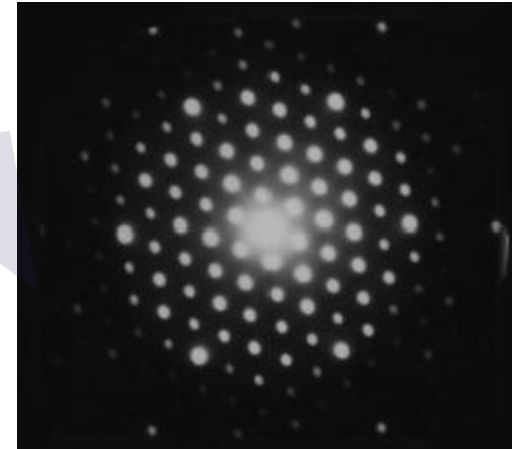
S. T. Misture and R. L. Snyder in
Encyclopedia of Materials: Science and Technology
K. H. Jürgen Buschow, et al. (Eds) Elsevier 2007



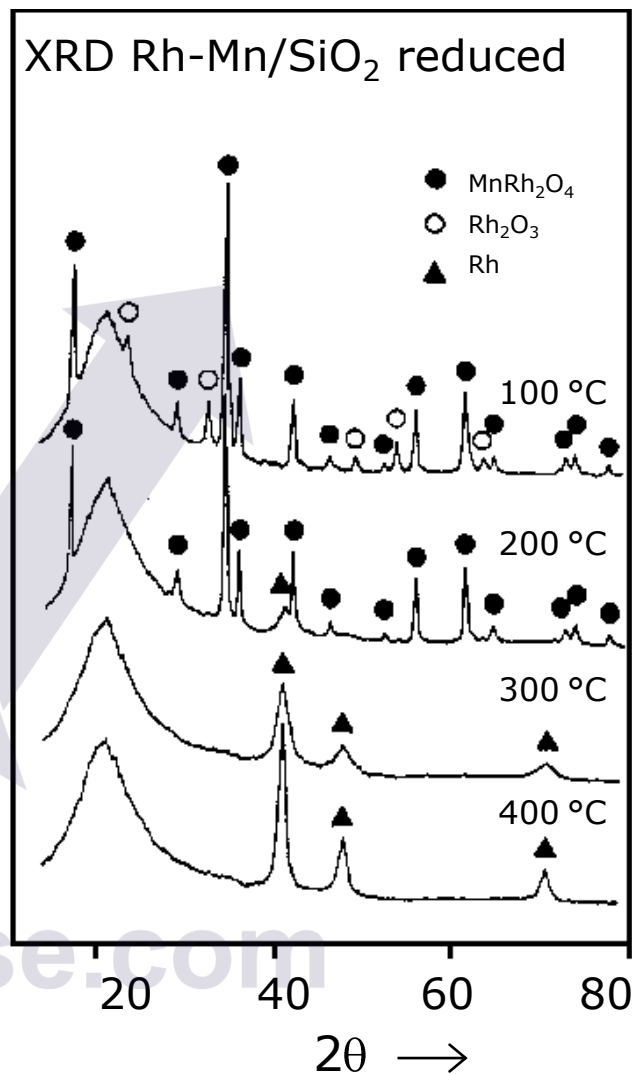
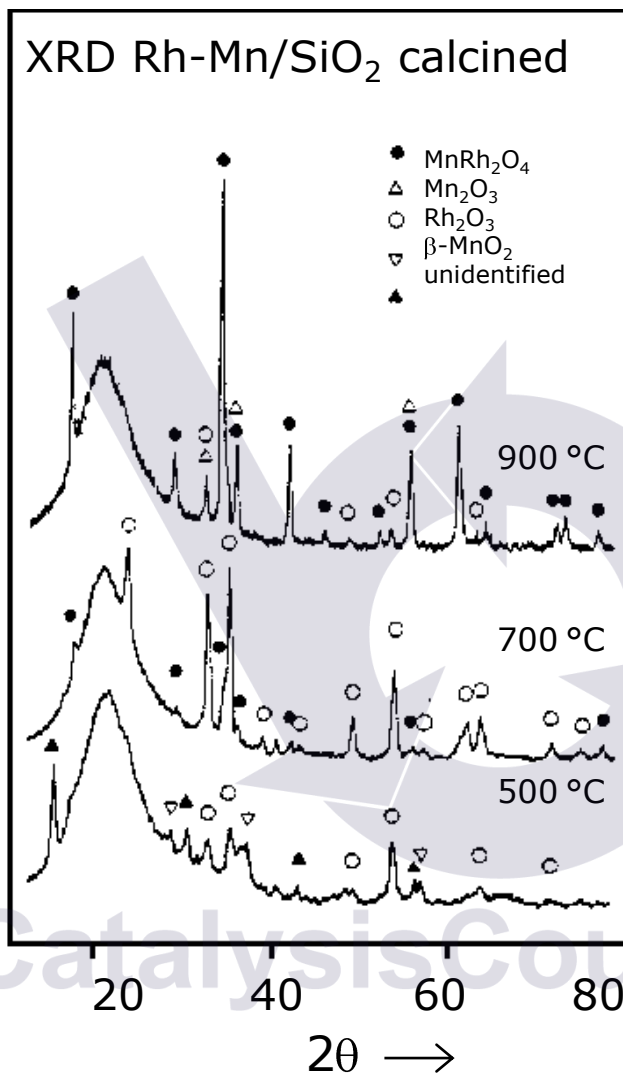
What do we learn from XRD?

From diffraction patterns we can:

- measure the spacings between layers of atoms, and thus identify phases
- determine the orientation of a single crystal or grain
- find the crystal structure of an unknown material
- measure the size, shape and internal stress of small crystalline regions



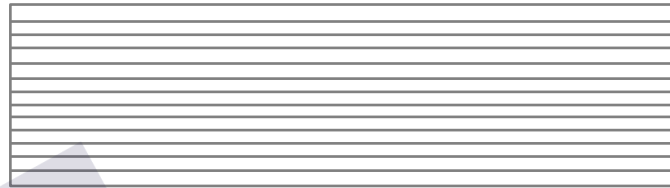
XRD: Identification of Phases in Rhodium Catalysts



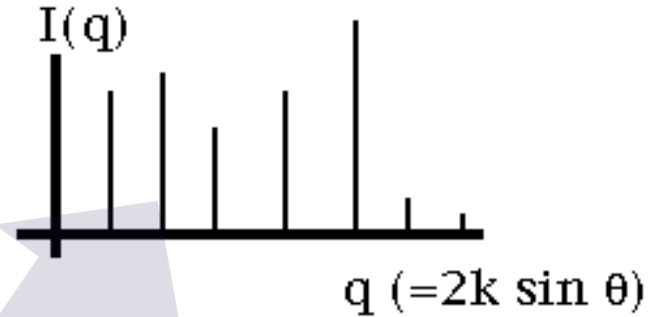
by comparing to standard diffraction data (ASTM)

XRD: linewidth and 'coherence length'

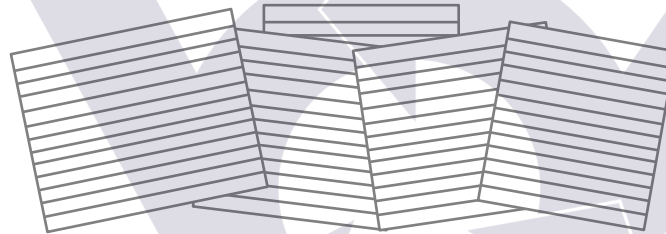
single crystal,
large crystalline
particles



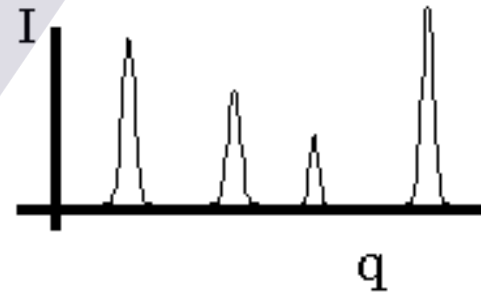
perfect long-range order



imperfect crystals,
nano particles

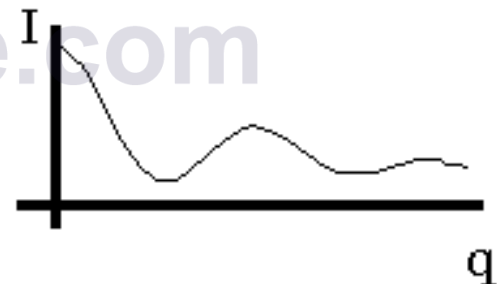


domains of medium-range order



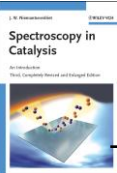
amorphous compounds
glasses, liquids

short-range order only



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XRD: X-ray Diffraction

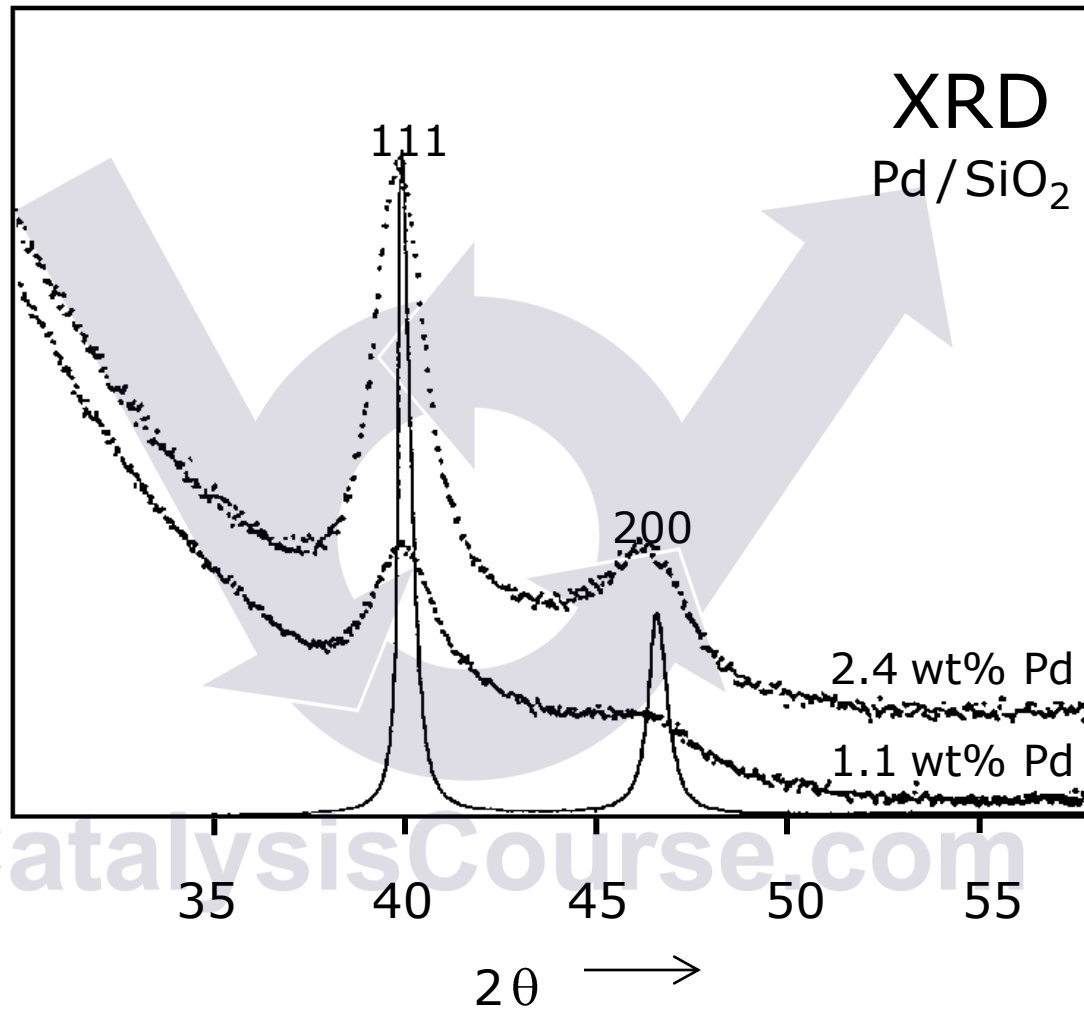


Features of Diffraction (Electron, X-ray, or Neutron)

- For a known structure, pattern can be calculated exactly.
- Symmetry of the diffraction pattern given by symmetry of the lattice.
- Intensities of spots determined by basis of atoms at each lattice point.
- Sharpness and shape of spots determined by perfection of crystal.
- Liquids, glasses, and other disordered materials produce broad fuzzy rings instead of sharp spots.
- Defects and disorder in crystals also result in diffuse scattering.

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XRD Line-broadening



Particle size estimate possible

Particle size from XRD line broadening

The Scherrer formula relates crystal size to line width:

$$\langle L \rangle = \frac{K \lambda}{\beta \cos \theta}$$



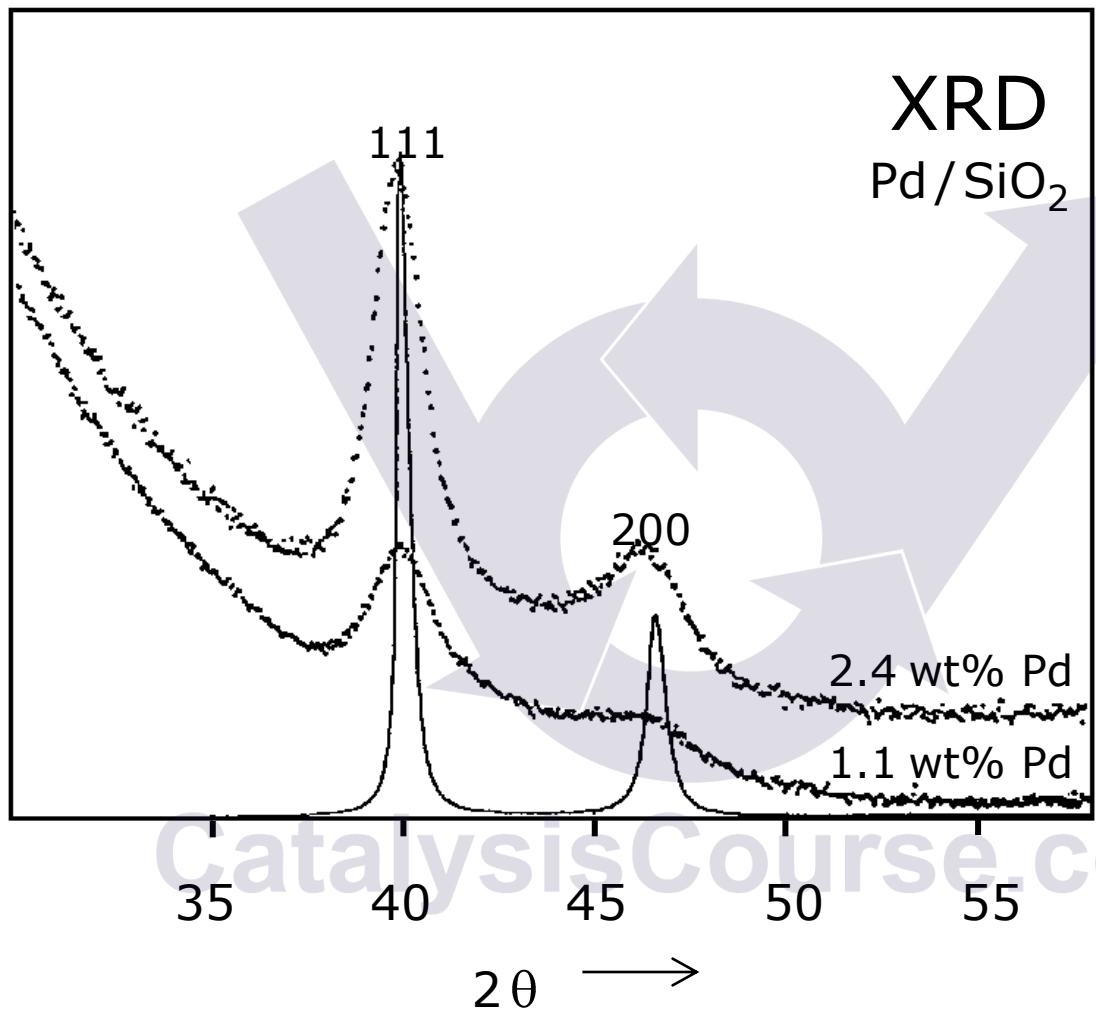
Paul Scherrer
1890-1969

in which

- $\langle L \rangle$ is a measure for the dimension of the particle in the direction perpendicular to the reflecting plane
- λ is the X-ray wavelength
- β is the peak width
- θ is the angle between the beam and the normal on the reflecting plane
- K is a constant (often taken as 1).

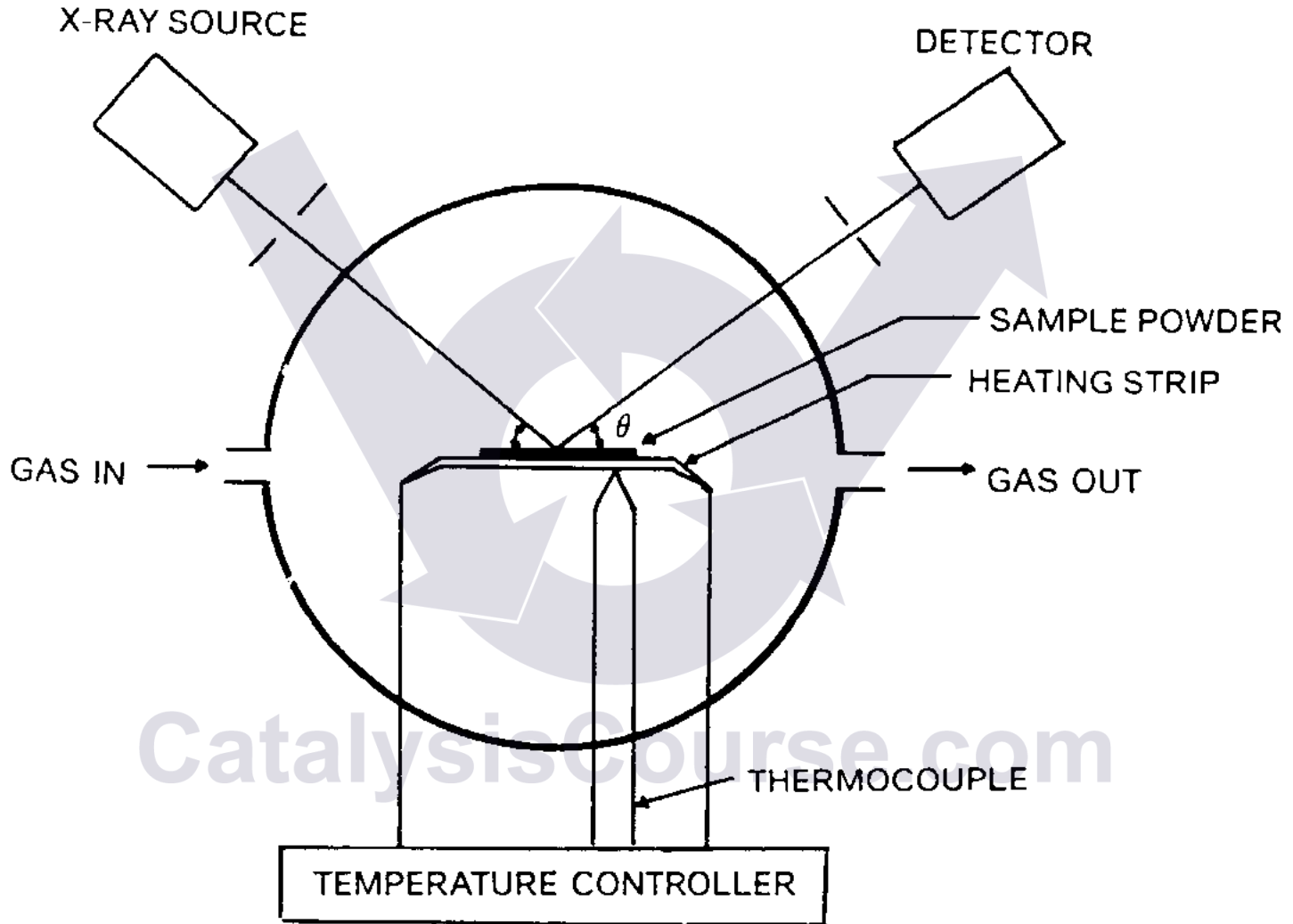
Broadening becomes smaller for smaller wavelength;
Mo K α (17.44 keV; 0.07 nm) versus Cu K α (8.04 keV; 0.15 nm)

XRD Line-broadening

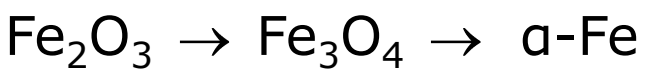
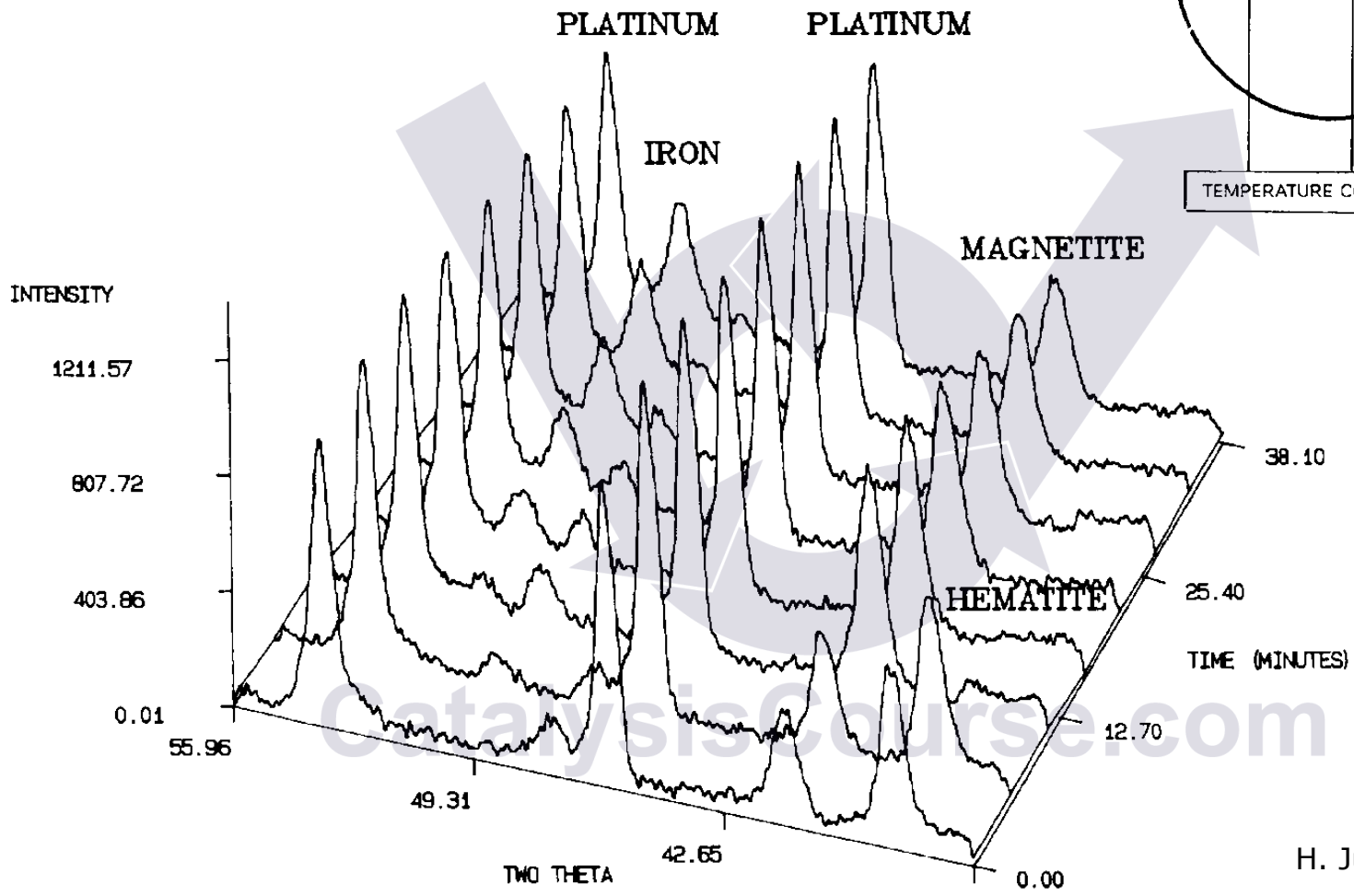
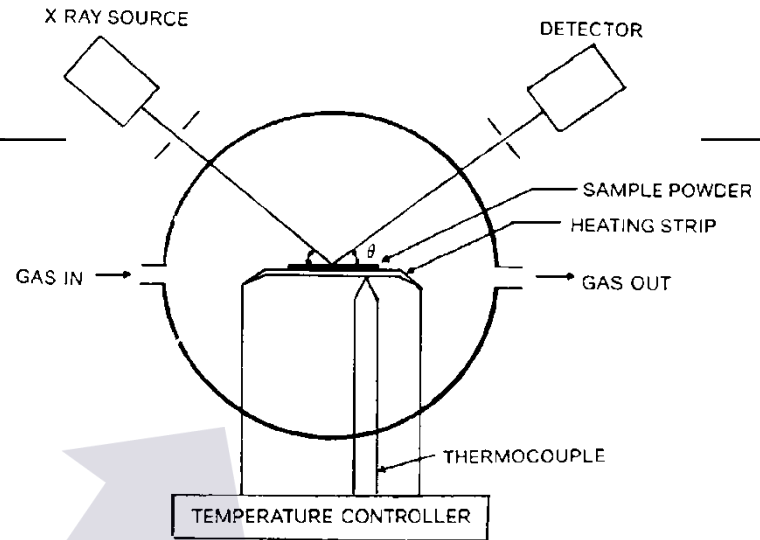


Scherrer equation:
~4.2 nm
~2.5 nm

in situ XRD

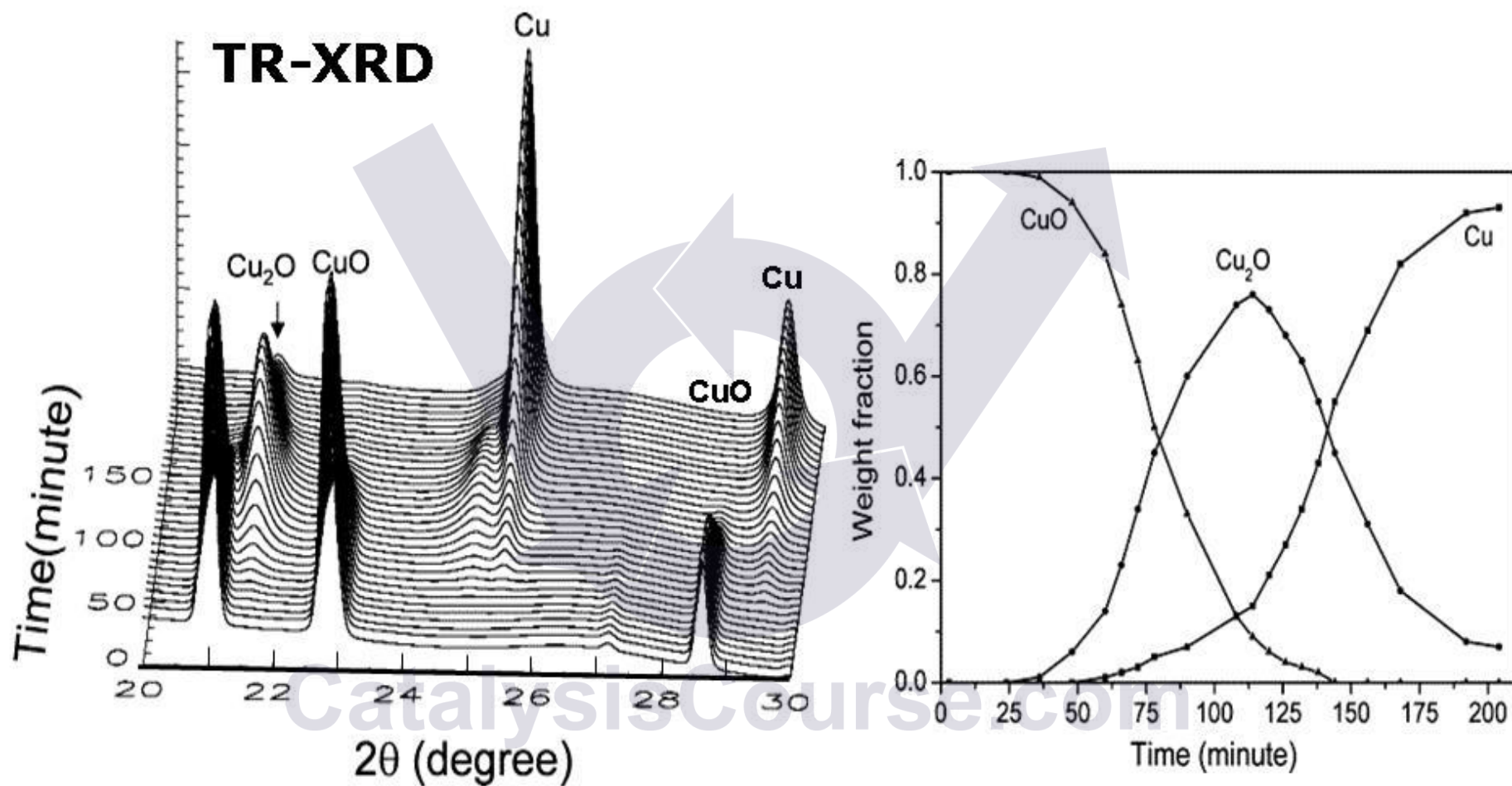


in situ XRD



H. Jung and W.J. Thomson,
J. Catal. 128 (1991) 218

XRD: Reduction of CuO by CO

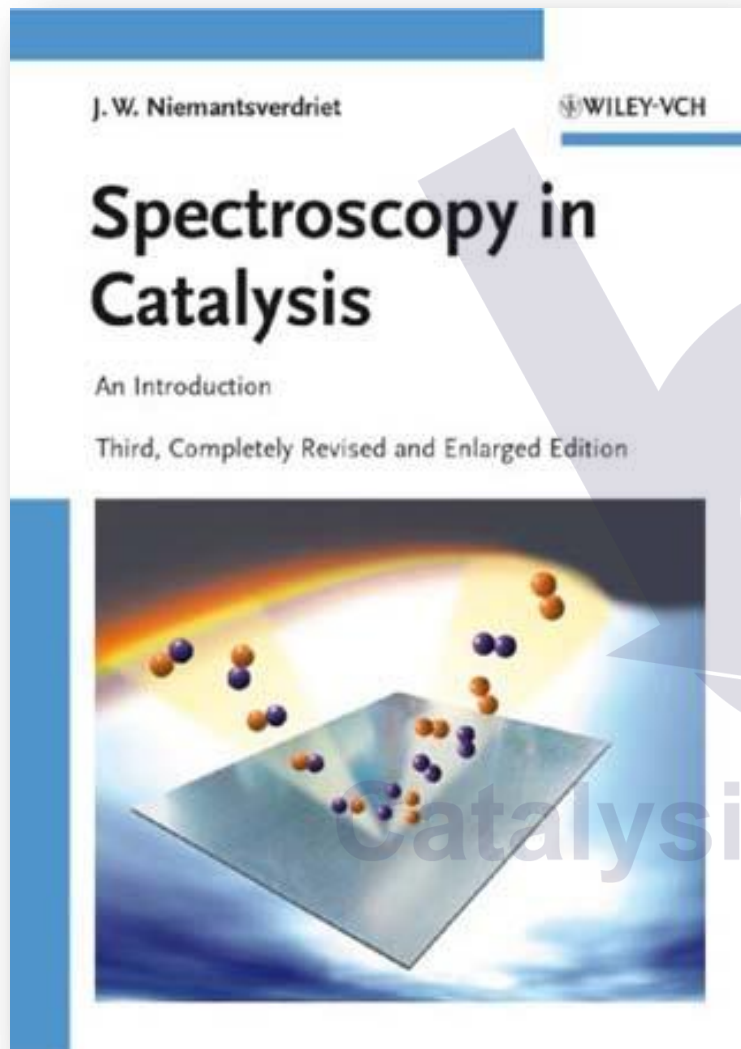


Conclusion

XRD:

- Phase identification
- Particle size estimate from line broadening
- In situ studies
- Careful:
 - only crystalline phases are detected
 - surface region is invisible

Download the handout for this lecture from
www.catalysiscourse.com



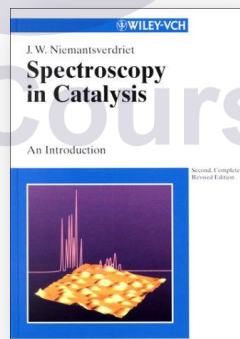
**Read more about X-ray
diffraction**

in Chapter 6 of

**Spectroscopy in Catalysis: An
Introduction, Third Edition**

J. W. Niemantsverdriet

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ISBN: 978-3-527-31651-9**



Version 2000

**gives many examples and
references to the literature**

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